

## Imidazolopiperazines: Lead Optimization of the Second-Generation Antimalarial Agents

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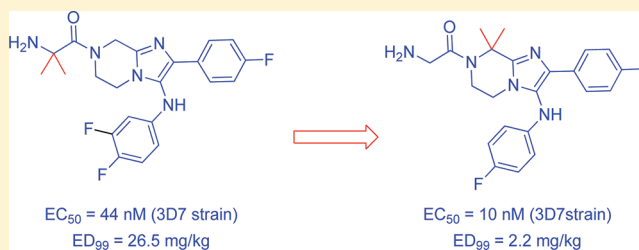
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### S Supporting Information

**ABSTRACT:** On the basis of the initial success of optimization of a novel series of imidazolopiperazines, a second generation of compounds involving changes in the core piperazine ring was synthesized to improve antimalarial properties. These changes were carried out to further improve the potency and metabolic stability of the compounds by leveraging the outcome of a set of *in vitro* metabolic identification studies. The optimized 8,8-dimethyl imidazolopiperazine analogues exhibited improved potency, *in vitro* metabolic stability profile and, as a result, enhanced oral exposure *in vivo* in mice. The optimized compounds were found to be more efficacious than the current antimalarials in a malaria mouse model. They exhibit moderate oral exposure in rat pharmacokinetic studies to achieve sufficient multiples of the oral exposure at the efficacious dose in toxicology studies.



## INTRODUCTION

After decades of neglect, the search for novel, effective antimalarials has been revived primarily due to emergence of drug resistance.<sup>1</sup> Widespread resistance has been reported not only for 8-aminoquinolines but recent reports indicate the appearance of increased parasite tolerance to artemisinin analogues, potentially predicating clinical resistance.<sup>2</sup> The absence of highly efficacious malaria vaccines highlights the importance of developing small-molecule drugs.<sup>3</sup> With support from the Wellcome Trust and Medicines for Malaria Venture (MMV), the NGBS consortium (the Novartis Institute for Tropical Diseases (NITD), the Genomics Institute of the Novartis Research Foundation (GNF), the Biomedical Primate Research Centre (BPRC), and the Swiss Tropical and Public Health Institute (STPHI)) was formed and set up to discover new antimalarials which are effective against multidrug resistant parasite strains. This collaboration has produced promising new leads.<sup>1a,4</sup> Here we report our continued effort to optimize a series of imidazolopiperazine compounds in search of a pre-clinical candidate. The lead compound we identified was compound **1**.<sup>5</sup> As described previously, an extensive SAR study on

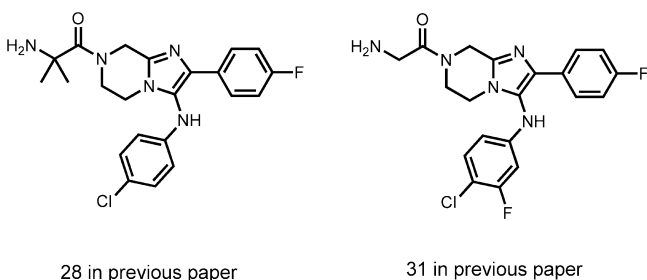
the three peripheral parts had yielded decent *in vitro* and *in vivo* antimalarial activities as well as favorable physicochemical and pharmacokinetic properties.

Our efforts toward the first-generation compounds led to candidates (like compound **1**) with double-digit nanomolar potency and moderate oral exposure in mice. However, most potent compounds with an unsubstituted piperazine had poor oral exposure which resulted in inferior *in vivo* efficacy when administered orally, for example, compounds **28** and **31** (Scheme 1) in the previous communication (Figure 1).<sup>5</sup> We were intrigued by the possibility of further refining the core structure of the molecule to further improve the properties. Our hypothesis was further reinforced by chemical stability and mouse *in vitro* metabolic stability results which led to the identification of two biologically inactive metabolites (**1a** and **1b**) arising from the oxidation of piperazine ring. Our plan was to substitute three possible positions (positions 5, 6, and 8) to

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## Scheme 1. Unsubstituted Piperazines Previously Disclosed



evaluate the corresponding effects on potency as well as a means to improve oral exposure.

## CHEMISTRY

On the basis of the designed molecules, it was clear that previous synthetic routes were not amenable for the desired compounds because we could not utilize the 3 + 2 cyclization.<sup>6</sup> Distinct syntheses were designed and executed to obtain substitution at different positions of a piperazine ring. Scheme 2 summarizes how the 8-substituted piperazines were prepared. The synthesis started with an alkylation reaction between

Cbz-protected glycine derivative **2** and bromide **3**. Adduct **4** was then refluxed with  $\text{NH}_4\text{OAc}$  in toluene to give imidazole **5** with removal of water using a Dean–Stark apparatus.<sup>7</sup> Regioselective *N*-alkylation with ethyl 2-bromoacetate followed by a one-pot deprotection/cyclization sequence furnished the lactam core, which was reduced by borane to furnish amine **8**. The quaternary  $\beta$ -carbon did not pose a significant amount of steric hindrance for the HATU mediated amidation reaction with *N*-Boc-glycine. Amide **9** was obtained at room temperature in decent yield and further was brominated to provide intermediate **10**. Then a standard Buchwald–Hartwig amination reaction with substituted aniline followed by deprotection finished the synthesis of compound **12** with the desired alkyl modification at position 8. The overall sequence involved nine steps with an overall yield of 17%.

Scheme 3 describes the synthesis of 6,6-dimethyl analogues. Analogous chemistry to compound **12** was used for the first two steps to provide imidazole **5a**. The compound was regioselectively *N*-alkylated with methyl chloride to obtain the key intermediate for the piperazine cyclization step. The intramolecular hydroamination reaction proved to be difficult, and after extensive screening, treating the substrate **5a** with a 5:1 mixture of acetic acid and methanesulfonic acid at 210 °C produced the cyclized core with simultaneous loss of the

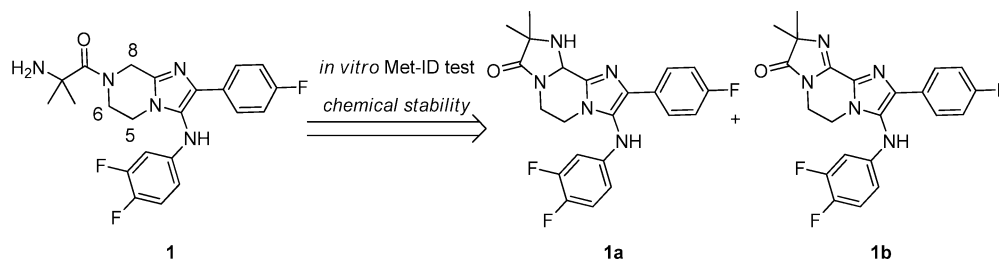
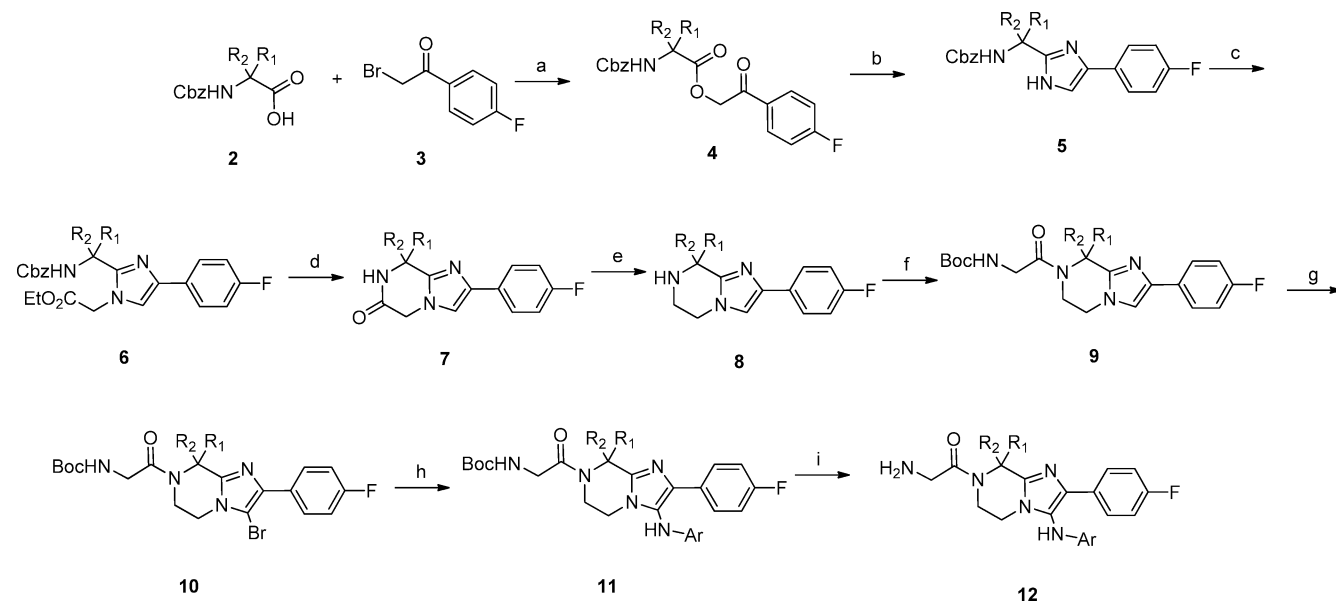
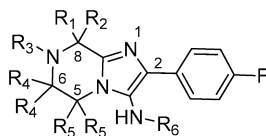


Figure 1. Central piperazine ring oxidation is a major metabolic and chemical instability site in compound **1**.

Scheme 2. Modification on the 8 Position of Imidazolopiperazine Core<sup>a</sup>

<sup>a</sup>Reagents and conditions: (a)  $\text{K}_2\text{CO}_3$ , DMF, rt, 84%; (b)  $\text{NH}_4\text{OAc}$ , toluene, reflux, 88% (c) ethyl 2-bromoacetate,  $\text{Cs}_2\text{CO}_3$ , DMF, rt, 83%; (d) Pd/C,  $\text{H}_2$  (balloon), MeOH, rt, 91%; (e)  $\text{BH}_3\cdot\text{THF}$ , THF, reflux, 95%; (f) HATU, DIEA,  $\text{CH}_2\text{Cl}_2$ , rt, 70%; (g)  $\text{Br}_2$ , AcOH,  $\text{CH}_2\text{Cl}_2$ , rt, 100%; (h)  $\text{ArNH}_2$ ,  $\text{Pd}_2(\text{dba})_3$ , xantphos,  $\text{Cs}_2\text{CO}_3$ , dioxane, 150 °C, 89% (for example **22**, Ar = 4F-Ph and  $\text{R}_1 = \text{R}_2 = \text{Me}$ ); (i) TFA,  $\text{CH}_2\text{Cl}_2$ , rt, 52% (for example **22**, Ar = 4F-Ph and  $\text{R}_1 = \text{R}_2 = \text{Me}$ ).

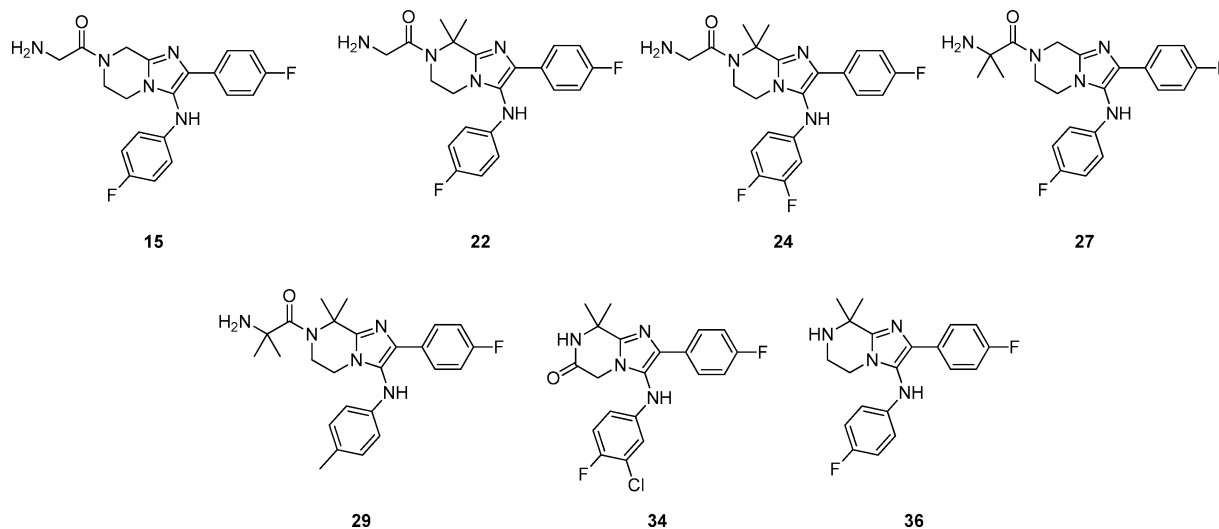


Table 1. SAR Study for Analogues with Substitution on the Piperazine Core<sup>a</sup>

compd	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	R <sub>4</sub>	R <sub>5</sub>	R <sub>6</sub>	<i>P. falciparum</i> strain EC <sub>50</sub>	
							3D7 (nM)	W2 (nM)
15	H	H	gly	H	H	4-FPh	24 ± 2	23 ± 8
16	H	H	gly	H	H	3Cl,4F-Ph	4 ± 3	4 ± 2
17	(S)-Me	H	gly	H	H	3Cl,4F-Ph	29 ± 2	24 ± 1
18	(R)-Me	H	gly	H	H	3Cl,4F-Ph	31 ± 4	29 ± 3
19	(S)-i-Pr	H	gly	H	H	3Cl,4F-Ph	343 ± 59	470 ± 130
20	(R)-i-Pr	H	gly	H	H	3Cl,4F-Ph	168 ± 15	202 ± 38
21	Me	Me	gly	H	H	3Cl,4F-Ph	27 ± 1	24 ± 1
22	Me	Me	gly	H	H	4-FPh	10 ± 3	6 ± 1
23	Me	Me	gly	H	H	4-MePh	8 ± 2	6 ± 1
24	Me	Me	gly	H	H	3,4-diFPh	4 ± 1	5 ± 1
25	H	H	gly	Me	H	4-FPh	82 ± 7	111 ± 17
26	H	H	gly	H	Me	4-FPh	80 ± 5	76 ± 15
27	H	H	$\alpha$ -Me-Ala	H	H	4-FPh	20 ± 5	25 ± 2
28	Me	Me	$\alpha$ -Me-Ala	H	H	4-FPh	10 ± 1	9 ± 1
29	Me	Me	$\alpha$ -Me-Ala	H	H	4-MePh	2 ± 1	4 ± 1
30	Me	Me	$\alpha$ -Me-Ala	H	H	3,4-diFPh	13 ± 3	7 ± 1

<sup>a</sup>Values are means of two experiments. Each assay plate has mefloquine, sulfadoxine, and artemisinin as internal standards. The EC<sub>50</sub> values for standard compounds match the literature values.

## Scheme 5. Structures of Key Compounds

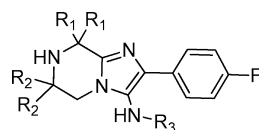


of the piperazine nitrogen could be modulated without complete loss in antimalarial activity.

**ADME and Pharmacokinetic Properties.** In parallel to the SAR studies to optimize the potency of the compounds series, we evaluated the in vitro ADME and in vivo pharmacokinetic properties of key analogues. In general, this series of compounds had very good Lipinski's rule-of-five compliance and most of the analogues were highly soluble (>175  $\mu$ M at pH 6.8), with the neutral lactams being least soluble among all the compounds. Table 3 summarizes the in vitro ADME profiles of the important compounds. Typically, a range of values in the permeability assays was observed. In the Caco-2 assays, the core piperazine compounds (both lactams as well as nonamide piperazine compounds) exhibited high apical to basal values

(31, 32, 34, and 35). The permeability of the compounds was reduced once the side chain amino acid was installed, but metabolic stability is improved (28 vs 35). There was no clear correlation between Caco-2 and PAMPA assay. However, the % absorbed was lower in the PAMPA assay for compounds with poor permeability in the Caco-2 assay. Upon comparing the calculated log *P* values with PAMPA as well as Caco-2 values, we discovered no correlation with any of the permeability parameters. Unsurprisingly, log *P* of 4.5 or higher was needed for having excellent percentage absorbed values in PAMPA assay (see Supporting Information). Removing the basicity of the piperazine ring by introduction of carbonyl group improved the mouse and human microsome stability of the compound, presumably by blocking the ethylene bridge (31 vs 32) but is

Table 2. SAR Study for Analogues with Substitution on the Piperazine Core



compd	R <sub>1</sub>	R <sub>2</sub>	R <sub>3</sub>	mouse microsome hepatic extraction ratio	<i>P. falciparum</i> strain EC <sub>50</sub>	
					3D7 (nM)	W2 (nM)
31	H	H	4-MePh	0.80	161 ± 13	178 ± 12
32	H	=O	4-MePh	0.70	1079 ± 285	1168 ± 188
33	Me	=O	4-MePh	0.72	138 ± 10	136 ± 20
34	Me	=O	3Cl,4F-Ph	0.61	36 ± 2	65 ± 3
35	Me	H	4-MePh	0.77	13 ± 2	13 ± 1
36	Me	H	4-FPh	0.80	5 ± 1	5 ± 1

Table 3. In Vitro ADME of Selected Analogues

entry	PAMPA <sup>a</sup>	Caco-2 <sup>b</sup> A-B	Caco-2 <sup>b</sup> B-A	CYP Met CL-Ms <sup>c</sup> (ER)	CYP Met CL-rat (ER)	CYP Met CL-Hu <sup>d</sup> (ER)	hERG <sup>e</sup> IC <sub>50</sub> (μM)
15	52	0.70	2.55	0.47	nt	<0.21	nt
22	51.8	1.26	2.99	0.48	0.66	0.49	13.4
24	51.5	0.61	1.3	0.46	0.69	0.53	>30
27	81	3.05	8.38	0.51	0.63	0.52	5.6
28	99.5	1.2	0.97	0.4	0.89	0.55	21.9
29	99.5	3.5	2.37	0.41	0.79	0.47	>30
31	99.5	8.2	2.36	0.87	0.96	0.79	nt
32	80.9	15	14.37	0.66	0.95	0.44	nt
33	nt	nt	nt	0.72	nt	nt	nt
34	97.1	3.22	3.35	0.61	0.68	0.72	>30
35	99.9	15.56	6.96	0.87	0.93	0.88	nt
36	99.9	nt	nt	0.8	0.92	0.92	nt

<sup>a</sup>Measured as percentage absorbed. <sup>b</sup>units = (cm/s × 10<sup>6</sup>). <sup>c</sup>Ms = mouse. <sup>d</sup>Hu = human. <sup>e</sup>Q-patch assay; nt = not tested.

Table 4. Complete in Vivo Pharmacokinetic Profile of Selected Analogues in Mice<sup>a</sup>

entry	IV PK parameters				PO PK parameters					
	V <sub>d</sub> (L/kg)	CL (mL/min/kg)	T <sub>1/2</sub> (h)	AUC <sub>(0-24 h)</sub> (h·nM)	C <sub>max</sub> (nM)	T <sub>max</sub> (h)	T <sub>1/2</sub> (h)	AUC <sub>(0-24 h)</sub> (h·nM)	F (%)	
15	12.3	188.9	2.7	1072	235	2.0	4.20	965	15	
22	10.2	49.2	2.0	4119	1538	1.0	4.35	12155	74	
24	7.5	27.5	5.5	7062	1723	3.0	5.21	21058	75	
27	23.7	60.0	6.2	3442	1803	1.7	8.95	17894	87	
29	7.5	36.1	18.0	7607	1642	1.0	14.54	32409	107	
34	2.7	12.5	2.8	16525	3651	1.0	2.71	50619	77	
36	1.2	25.9	0.8	9090	6404	1.0	2.95	19447	53	

<sup>a</sup>PO dosing; dose (mg/kg), 20; species/strain, mouse, male balb/c; formulation, 2.5 mg/mL in PEG300/D5W, 3:1, solution; V<sub>d</sub> = volume of distribution; CL = clearance; C<sub>max</sub> = maximal concentration; T<sub>max</sub> = time at maximal concentration; AUC = area under the curve; F = oral bioavailability.

not evident in rat microsomes. Introduction of the dimethyl group on the 8-position of the piperazine ring does not improve overall metabolic stability of the compound (15 vs 28 or 31 vs 35) despite the metabolite identification results presented earlier. This may be due to the ethylene bridge in the piperazine ring representing a metabolic soft spot and enabling metabolic switching. In general, this scaffold exhibited poor rat metabolic stability when compared to other species. One interesting finding with the substituted piperazine analogues was the improvement of activity against the hERG channel as determined in the automated patch clamp assay. Compared to the unsubstituted piperazine analogue 27, we were able to decrease inhibition of the hERG channel by at least 2–3-fold, with most

analogues (such as 22) suggesting a lower potential for QT interval prolongation in compounds with a substituted piperazine.

**Pharmacokinetic Properties.** Compounds series found to have favorable potency and in vitro ADME properties were tested in snapshot PK, an abbreviated PK study shown to accurately bin compounds into low, moderate, or high oral exposure.<sup>9</sup> Compounds predicted to have reasonable oral exposure would be tested in mice using both PO and IV routes of administration (Table 4). The in vivo clearance generally was low to moderate and significantly improved when compared to the corresponding unsubstituted analogues (15, 22). This resulted in higher maximum concentration as well as



improvement in AUC values in the oral arms as well. Lactam compound **34** exhibited highest oral AUC than all other analogues, and it also had lower intrinsic clearance with moderate volume of distribution. To test whether red cell partitioning was occurring to explain the high volume of distribution for the basic compounds, we carried out an *in vitro* experiment in which compound **22** was incubated in naïve rat whole blood as well as plasma for 30 min. After centrifuging the blood and quantifying the amount of compound in the red blood cells, we determined that the concentration of the compound was 3.47 times higher in whole blood than in plasma alone, indicating that the compound is either adhering to the RBC's or getting distributed inside the cells.

**In Vivo Efficacy.** Because we had examples of compounds with different modifications with acceptable bioavailability, we decided to carry on the efficacy experiment in mice. The *in vivo* antimalarial activities of the optimized compounds were evaluated using a *Plasmodium berghei* mouse survival model.<sup>8</sup> In this test, groups of three *P. berghei* infected mice were treated orally with compound one day after infection. The key readouts were the percentage of parasitemia reduction on day 3 postinfection compared to the untreated mice and the prolongation of survival of the infected mice.

Table 5 summarizes the *in vivo* efficacy test results of compound in comparison with standard antimalarials chloroquine

**Table 5. In Vivo Efficacy of Compounds on *Plasmodium berghei* Mouse Survival Model Following Different Dose Levels in Comparison with Known Antimalarials<sup>a</sup>**

entry	1 × 30 mg/kg		1 × 100 mg/kg		3 × 30 mg/kg	
	activity (%)	survival (day)	activity (%)	survival (day)	activity (%)	survival (day)
22	99.5	16.3	99.4	14.0	99.8	17.7
24	99.0	15.0	99.1	16.7	99.7	17.0
29	99.0	15.3	99.0	16.7	99.7	18.0
34	47.0	6.3	97.0	6.7	66.0	7.0
36	90.0	7.0	99.3	9.7	99.7	12.0
CQ <sup>b</sup>	99.5	9.0	99.6	12.5	99.8	13.6
AS <sup>b</sup>	95.6	5.8	98.0	7.0	99.0	12.2

<sup>a</sup>Activity = average parasitemia reduction; survival = average lifespan after infection (6–7 days for untreated control mice); compounds were formulated in 75% PEG300/25% D5W (5% dextrose in water); CQ, chloroquine; AS, artesunate. <sup>b</sup>10% ethanol, 30% PEG400, 6% vitamin ETPGS.

(CQ) and artesunate (AS). It was observed that compounds **22**, **24**, and **29** exhibited >99% parasitemia reduction at a low dose of 30 mg/kg with a modest survival of 15 days or higher. Increasing the dose to 100 mg/kg or dosing the compound 3 × 30 mg/kg did not lead to increase in survival. Currently, we are investigating the possibility of recrudescence of parasites or development of resistance for the compound *in vivo* to better explain the lack of curative activity in this model. We decided to determine the ED<sub>99</sub> representative examples for the substituted and nonsubstituted piperazine cores and found the substituted piperazine core was ~10-fold more efficacious (compound **22** ED<sub>99</sub> = 2.2 mg/kg) than the unsubstituted core (compound **1** ED<sub>99</sub> = 26.5 mg/kg). Compounds **34** and **36** clearly proved inferior both in terms of parasitemia reduction as well as survival in mice, indicating the need for the amino acid side chain in providing better mouse efficacy.

On the basis of the efficacy results, we decided to study the oral exposure of several compounds in rats. Rat was chosen as

the second animal species for oral exposure evaluation as it may serve as the *in vivo* preclinical toxicity species. Taking into account the nonoptimal metabolic stability data for rats for this scaffold, we decided to study the oral exposure of the compound in rat for a period of 5 h in the oral arm only to get an estimation of the exposure (Table 6).<sup>9</sup> The results show that

**Table 6. Rat Snapshot PK Results for Interesting Compounds<sup>a</sup>**

entry	PK parameters		
	AUC <sub>(0–5 h)</sub> (h·nM)	C <sub>max</sub> (nM)	AUC <sub>(0–5 h)</sub> /dose [(min·μg/mL)/(mg/kg)]
22	487	126	1.2
24	367	96.66	0.95
29	NA	BLQ	
34	2097	504.44	5.07
36	423	202	0.9

<sup>a</sup>PO dosing; dose (mg/kg), 20; species/strain, rat, Wistar; formulation, 2.5 mg/mL in PEG300/D5W, 3:1, solution; BLQ, below level of quantification; NA = could not be determined.

the 8-substituted piperazine compounds had inferior exposure.  $\alpha$ -Methylalanine subclass of compounds like **29** did not exhibit any *in vivo* exposure and were not further pursued. In line with the better rat metabolic stability (ER = 0.68), nonbasic lactam compound **34** demonstrated the best rat exposure despite lower mouse efficacy.

As the oral exposure of compounds **22** was low in rats but demonstrated superior mouse efficacy, we decided to dose compound **22** at higher doses to make sure that sufficient multiples of exposure of the mouse efficacious dose could be achieved (Table 7). We observe that when the dose was increased to 30 mg/kg from 10 mg/kg, there was overproportional 7-fold increase in AUC as well as 2-fold increase in bioavailability. Increasing the dose further to 100 mg/kg dose with suspension formulation led to 55-fold increase in AUC with a 3-fold increase in oral bioavailability. The AUC at the mouse efficacious dose (2 mg/kg) was 780 h·nM, therefore a ~45× exposure multiple is achieved at 100 mg/kg in the rat. The multiple of exposure provides a sufficient window to use the rat as a rodent toxicology species. It is unclear at this time if this overproportional exposure at higher doses is due to saturating metabolism or an efflux transporter, but given the partial efflux observed in Caco-2, this may explain this multiple dose exposure results.

**In Vitro Toxicology.** Compound **22** was subjected to a series of assays to evaluate the toxicity profile. This drug candidate was inactive on a panel of human cytochromes P450 (IC<sub>50</sub>s are greater than 6 μM for CYP1A2, CYP2C19, CYP2C9, CYP2D6, and CYP3A4), and the observed cytotoxicity CC<sub>50</sub> were all greater than 12 μM on a panel of mammalian cell lines (293T, Ba/F3, CHO, HEp2, HeLa, Huh7), which translates to an adequate selectivity index (SI > 100). Compound **20** was found negative in the mini-Ames and micronucleus tests, which indicated a low mutagenic potential. Finally, considering the efficacious plasma C<sub>max</sub> in the mouse at 2 mg/kg is low (~75 nM), the risk of cardiotoxicity is considered to be minimal where the hERG channel inhibitory activity IC<sub>50</sub> = 13.4 μM, providing a greater than 170-fold window.

Table 7. In Vivo Rat Dose Escalation PK Tests for Compound 22

entry	IV PK parameters				PO PK parameters					
	$V_d$ (L/kg)	CL (mL/min/kg)	$T_{1/2}$ (h)	AUC (h·nM)	dose mg/kg	$C_{max}$ (nM)	$T_{max}$ (h)	$T_{1/2}$ (h)	AUC <sub>(all)</sub> (h·nM)	F (%)
22	13.7	67.5	4.6	1831	10	91	1.5	4.7	974	20
					30	580	4.3	6.3	7337	40
					100	34885	8.0	8.4	34885	57

## SUMMARY

Imidazolopiperazine compounds had been proven to be effective against multidrug resistant parasite strains with minimum toxicity to host-cell lines. However, early lead compound such as compound 15 and 27 still suffer from moderate potencies, relatively poor rat PK, and potent hERG inhibitory activity. Directed by the identification of in vitro metabolites formed by compound 1, the piperazine core of 15 was systematically modified by a gem-dimethyl functionality at all three possible places. The 8,8-dimethyl analogues demonstrated excellent potency as well as improved oral exposure in the mouse. This also allowed us to use the glycine amide that reduced our hERG risk when compared to 27. The first generation of compounds exhibited efficacy that was comparable to the known antimalarials in the *P. berghei* mouse model, however, compound 22 was more efficacious than chloroquine and artesunate especially when parasitemia reduction is compared in terms of ED<sub>99</sub> values. Although compound 22 demonstrated poor rat PK at low doses, with higher doses it was able to achieve the necessary exposure multiples calculated based on the plasma concentration found to be efficacious in the mouse model. Compound 22 is thus currently under pre-clinical safety assessment in rats as a potential candidate for human trials.

## EXPERIMENTAL SECTION

**Experimental Protocols (Biology).** *Maintenance of Plasmodium falciparum Cultures.* *Plasmodium falciparum* cultures were grown in O<sup>+</sup> RBCs using culturing media: RPMI 1640 media (no phenol red) with L-glutamine and containing 0.05 mg/mL gentamicin, 0.014 mg/mL hypoxanthine, 38.4 mM Hepes, 0.20% sodium bicarbonate, 0.20% glucose, 3.4 nM NaOH, 5% human serum, and 1.25% albumax. Using traditional *P. falciparum* protocols published previously, 25 mL cultures were maintained in 75 cm<sup>2</sup> flasks (Fisher cat. no. 10-126-41) at 5% hematocrit and parasitemia ranging between 1% and 10%.<sup>10</sup> Once parasitemia reached 8–10%, cultures are diluted down to 1% or higher depending on needs. Maintenance required daily media changes and fresh blood every two weeks. Cultures were gassed for approximately 30 s to 1 min using a blood gas mixture to maintain a gas composition of 96% nitrogen, 3% carbon dioxide, and 1% oxygen and incubated at 37 °C. On day 1 of the assay, percent parasitemia was determined by obtaining a 1  $\mu$ L blood smear. The smear was fixed onto the slide by placing in methanol for 30 s, stained in 10% Geimsa stain, and the percent of parasitized erythrocytes vs uninfected erythrocytes was determined by microscopy with a light microscope.

*Antiplasmodial Proliferation Inhibition Assay: 384-Well Plate Format.* *P. falciparum* strains to be used for screening were prepared with screening media (culturing media without human serum but supplemented with 0.5% Albumax II) and fresh erythrocytes. Twenty  $\mu$ L of screening media were dispensed via the MicroFlo (BioTek) into 384-well, black clear-bottom assay plates ( $\mu$ Clear GNF custom plates by Griener Bio-One). Then 50 nL of compounds were transferred using a PlateMate Plus (Matrix) or the GNF Systems PinTool into the assay plates containing screening media to a final maximum concentration of 10  $\mu$ M in a 12 point dose–response (1/2 log serial

dilution). On the basis of the measured parasitemia levels, each parasite strain was diluted to yield 0.5% parasitemia, and 50% hematocrit blood (uninfected erythrocytes) was added to a concentration of 4.17%. This diluted culture was used in dispensing the remaining 60% of the assay volume (30  $\mu$ L) into the prewarmed assay plates via the MicroFlo liquid dispenser. The final parasitemia was 0.3% and final hematocrit was 2.5% in a total volume of 50  $\mu$ L. Assay plates were incubated at 37 °C for 72 h in a gas chamber filled with low oxygen blood gas. After the 72 h incubation (equivalent to 1–2 cycles during the blood stage), a prepared mixture of the lysis buffer (5 mM EDTA, 1.6% Triton X-100, 20 mM Tris-HCl, 0.16% saponin) in water and SYBR Green detection (0.1%) reagent was dispensed at 10  $\mu$ L per well using the MicroFlo. Cultures were incubated for an additional 24 h at 25 °C before measuring fluorescence intensity using the Envision plate reader (Perkin-Elmer) with a 505 dichroic mirror. Excitation and emission filters are 485 and 530 nm, respectively. Data were normalized based on maximum fluorescence signal values for DMSO treated wells (no inhibition by compound) and the minimum fluorescence signal values for wells containing the highest concentration of inhibitor control compounds, for example, pyrimethamine at a final concentration of 10  $\mu$ M. Data were analyzed on a plate-by-plate basis and compared to reference compounds that were always included on every plate, typically artemisinin, mefloquine, and pyrimethamine. EC<sub>50</sub> values were obtained using a custom curve fitting model, and a standard logistic regression model was applied for curve fitting. The quality of the assay run was assessed by the performance of the reference compounds where the EC<sub>50</sub> must be within 3-fold of the standard reference values for the assay plate to pass requisite data quality control needs. Additionally, all compounds were typically assayed at a minimum in duplicate (independent assay plates) and EC<sub>50</sub> values ideally must not vary by more than 2-fold between plates.

**Metabolic Stability.** The metabolic stability of drug candidates was determined in human, mouse, and rat liver microsomes using the compound depletion approach, quantified by LC/MS/MS. The assay measured the rate and extent of metabolism of chemical compounds by measuring the disappearance of the parent compound. The assay determined the compound's in vitro half-life ( $T_{1/2}$ ) and hepatic extraction ratios (ER) and predicted metabolic clearance in human, rat, and mouse species.<sup>11</sup>

**PAMPA Assay.** Permeation experiments were carried out in 96-well microtiter filter plates. The compound concentration was determined by UV absorption measured with a SpectraMax190 microplate spectrophotometer (Molecular Devices Corporation, Sunnyvale, CA, USA) at absorption wavelengths between 260 and 290 nm.<sup>12</sup>

**Caco-2 Assay.** The Caco-2 assay was carried out in a 96-well format, and compound concentrations in each chamber were measured by LC/MS. The assay offers apparent permeability data of test compounds from the apical to the basolateral chambers [ $P_{app}$  (A–B)] and from the basolateral to the apical chambers [ $P_{app}$  (B–A)]. The results can be used to predict the in vivo oral absorption (using  $P_{app}$  (A–B)) and the transport mechanism of the test compounds across GI membrane (ratio of  $P_{app}$  (B–A): $P_{app}$  (A–B) and log  $P$  lipophilicity data).<sup>13</sup>

**hERG (Q-Patch) Assay.** The assay utilizes electrophysiology measurement on the electric current passing through hERG channel that is heterologously expressed in a stable CHO cell line. Channels were open by a hERG-specific voltage protocol, and the compound effects were directly characterized by their block of the hERG current. The assay was performed on an automated platform, Q-Patch. All

compounds were tested by 4-point dose–response curves, and  $IC_{50}$  values (between 0.2 and 30  $\mu$ M) were measured.<sup>14</sup>

**In Vivo PK.** *In Vivo Pharmacokinetics Studies in Mice.* In-life studies were carried out under protocols approved by the Animal Care and Use Committee (IACUC) of GNF. All compounds were formulated at a concentration of 2.5 mg/mL in 75% PEG300 and 25% DSW (5% dextrose in distilled water) solution and were filtered using a 0.45  $\mu$ m syringe filter. The filtered solutions were dosed intravenously via the lateral tail vein at 5 mg/kg with a dosing volume of 2 mL/kg to male Balb/c mice ( $n = 3$  per group). The filtered solutions were also administered orally at 20 mg/kg (with a dosing volume of 8 mL/kg) to another group of mice ( $n = 3$  per group). Six blood samples of 50  $\mu$ L each were collected serially from each animal via retro orbital sinus up to 24 h after dosing. Blood samples were collected into heparin microtainer and centrifuged and plasma separated and frozen until analysis. Plasma samples were analyzed by LC-MS/MS.

*In Vivo Pharmacokinetics Studies in Rats (Snapshot and Full PK).* Compounds were formulated at a concentration of 2.5 mg/mL in 75% PEG300 and 25% DSW solution. In rat snapshot PK, two rats (Wistar rats) were administered orally at 10 mg/kg dose and four blood samples (100  $\mu$ L) were taken at 0.5, 1, 3, and 5 h post dosing. Before the sample analysis, the plasma samples were pooled based on the sampling time. In the rat full PK studies, the formulations were filtered before the administration to rats. Three rats were dosed intravenously (3 mg/kg), and three rats were administered orally (10 mg/kg). Six blood samples (100  $\mu$ L per sample) were collected serially from each rat within 24 h post dosing.

**Extraction and LCMS Analysis.** Plasma samples (20  $\mu$ L) were extracted with solution of acetonitrile:methanol (3:1) containing internal standard. The samples were vortexed and then centrifuged with an Eppendorf centrifuge 5810R (Eppendorf, Hamburg, Germany) at a setting of 4000 rpm for 5 min at 4 °C. The supernatant was transferred to a clean 96-well analysis plate for LC/MS/MS analysis. An aliquot (10  $\mu$ L) was injected onto a Phenomenex C18 guard column (4 mm  $\times$  2 mm) followed by a Zorbax SB-C8 analytical column (2.1 mm  $\times$  30 mm, 3.5  $\mu$ m, Agilent Technologies Inc., Palo Alto, CA, USA). Separation was carried out using a gradient elution method with a mobile phase consisting of 0.05% formic acid in water (solvent A) and 0.05% formic acid in acetonitrile (solvent B). The flow rate was 600  $\mu$ L/min. The HPLC system, consisting of Agilent 1100 series binary pump (Agilent Technologies Inc.), Agilent 1100 series micro vacuum degasser (Agilent Technologies Inc.), HTC PAL CTC analytics autosampler (LEAP Technologies, Carborro, NC), and VICI two-position actuator (Valco Int., Houston, TX, USA) was interfaced to a MDS SCIEX API-4000 triple quadrupole mass spectrometer (MDS Inc., Toronto, Canada). Mass spectral analyses were carried out using electrospray source in the positive ion mode and using multiple reaction monitoring (MRM) for quantification. MRM transition of compound 15 383.9/101.2, compound 22 412.2/312.1, compound 24 430.1/330.1, compound 27 412.2/327.2, compound 29 436.1/308.1, compound 34 430.0/346.1, and compound 36 355.0/272.3 was used with its optimized MS parameters, respectively.

**Pharmacokinetic Analysis.** The mean value from the three animals at each time point was plotted against time to give plasma concentration time profile. Pharmacokinetic parameters were determined using Watson LIMS, version 7.2 (Thermo Electron Corporation, PA, USA), by noncompartmental analysis. Pharmacokinetic parameters for compounds were calculated by noncompartmental regression analysis using an in-house fitting program developed at GNF. The oral bioavailability ( $F$ ) was calculated as the ratio between the area under the curve (AUC) following oral administration and the AUC following intravenous administration corrected for dose ( $F = AUC_{po} \times dose_{iv} / AUC_{iv} \times dose_{po}$ ).

**In Vivo Antimalarial Activity.** All in vivo efficacy studies were approved by the veterinary authorities of the Canton Basel-Stadt. In vivo antimalarial activity was usually assessed for groups of three female NMRI mice (20–22 g) intravenously infected on day zero with

$2 \times 10^7$  erythrocytes parasitized with *P. berghei* GFP ANKA malaria strain (PbGFPCON donation from A. P. Waters and C. J. Janse, Leiden University).<sup>8a,15</sup> Untreated control mice die typically between day six and day seven post infection. Experimental compounds were formulated in 75% PEG300/25% DSW as indicated. Chloroquine and artesunate were formulated in 10% ethanol, 30% PEG400, and 6% vitamin E TPGS. Compounds were administered orally in a volume of 10 mL/kg either as a single dose (24 h post infection) or as three consecutive daily doses (24, 48, and 72 h post infection). With the single dose regimen, parasitemia was determined (72 h post infection) and for the triple dose regimen (96 h post infection) using standard flow cytometry techniques.<sup>15</sup> Activity was calculated as the difference between the mean percent parasitemia for the control and treated groups expressed as a percentage of the control group. The survival time in days was also recorded up to 30 days after infection. A compound was considered curative if the animal survived to day 30 after infection with no detectable parasites.

**Experimental Protocols (Chemistry).** *Materials and Methods.* Unless otherwise noted, materials were obtained from commercial suppliers and were used without purification. Removal of solvent under reduced pressure refers to distillation using Büchi rotary evaporator attached to a vacuum pump ( $\sim 3$  mmHg). Products obtained as solids or high boiling oils were dried under vacuum ( $\sim 1$  mmHg). Purification of compounds by high pressure liquid chromatography was achieved using a Waters 2487 series with Ultra 120 5  $\mu$ m C18Q column with a linear gradient from 10% solvent A (acetonitrile with 0.035% trifluoroacetic acid) in solvent B (water with 0.05% trifluoroacetic acid) to 90% A in 7.5 min, followed by 2.5 min elution with 90% A.

<sup>1</sup>H NMR spectra were recorded on Bruker XWIN-NMR (400 or 600 MHz). Proton resonances are reported in parts per million (ppm) downfield from tetramethylsilane (TMS). <sup>1</sup>H NMR data are reported as multiplicity (s singlet, d doublet, t triplet, q quartet, quint quintet, sept septet, dd doublet of doublets, dt doublet of triplet, bs broad singlet), number of protons, and coupling constant in hertz. For spectra obtained in CDCl<sub>3</sub>, DMSO-*d*<sub>6</sub>, and CD<sub>3</sub>OD, the residual protons (7.27, 2.50, and 3.31 ppm, respectively) were used as the reference.

Analytical thin-layer chromatography (TLC) was performed on commercial silica plates (Merck 60-F 254, 0.25 mm thickness); compounds were visualized by UV light (254 nm). Flash chromatography was performed either by CombiFlash (separation system Sg. 100c, ISCO) or using silica gel (Merck Kieselgel 60, 230–400 mesh). The purity and quantitative analysis were determined with a Waters ZQ2000 LC/MS system, which employed an Acquity UPLC system (binary pump, column compartment, autosampler, and diode array UV detector). The eluent was split between the mass spectrometer and an Antek chemiluminescent nitrogen detector (CLND). The mobile phases used were (A) 95% H<sub>2</sub>O/5% MeOH/IPA (75/25, v/v) + 0.05% formic acid and (B) MeOH/IPA (75/25, v/v) + 0.035% formic acid. A gradient HPLC method with flow of 0.4 mL/min started at 2% B, with a hold of 1.0 min before a linear increase to 95% B at 3.50 min, with a 30 s hold. From 4.0–4.25 min, the mobile phases were returned to starting conditions. Total run time was 5.0 min. The column used was a Thermo Synchronis C18 2.1 mm  $\times$  30 mm, 3  $\mu$ m. The mass spectrometer was operated in positive mode, with a spray voltage of 3.2 kV and cone voltage of 30 V. The source and desolvation temperatures were 130 and 400 °C, respectively, with 600 L/h of nitrogen desolvation flow. All the final compounds reported in this communication had the purity of at least 95%. Elemental analyses were carried out by Midwest microlabs LLC, Indianapolis, IN, USA.

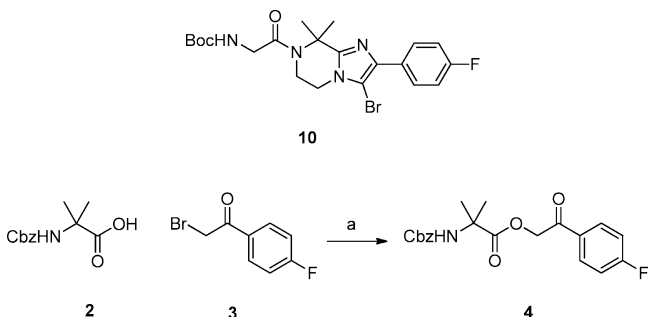
**Synthesis of 1a.** Compound 1a was prepared from 1 by the following method: to a stirred solution of 1 (12 mg, 0.03 mmol) was added MnO<sub>2</sub> (52 mg, 0.60 mmol). The reaction mixture was stirred at room temperature (rt) for 2 h. LC/MS test showed that 1 was almost consumed and the desired product ( $[M + 1] = 428$ ) was detected as one of the major peaks. Solid was filtered off, and solvent was



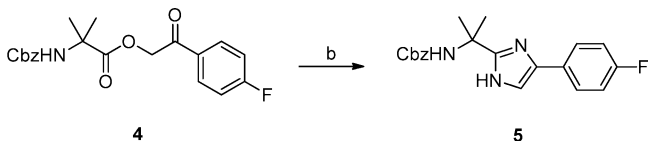
removed. The residue was subjected to a mass-triggered HPLC purification to give **1a** as well as **1b**.

$^1\text{H}$  NMR for **1a** (400 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  7.59–7.63 (m, 2H), 7.09 (t,  $J = 6.4$  Hz, 2H), 6.95–7.02 (m, 1H), 6.46–6.52 (m, 1H), 6.34–6.38 (m, 1H), 5.92 (s, 1H), 4.33 (dd,  $J = 4.4, 14$  Hz, 1H), 3.96 (dd,  $J = 4.8, 13.2$  Hz, 1H), 3.70–3.78 (m, 1H), 3.46–3.54 (m, 1H), 1.73 (s, 3H), 1.36 (s, 3H).  $m/z = 428.1$  ( $M + 1$ ).  $^1\text{H}$  NMR for **1b** (400 MHz,  $\text{CD}_3\text{OD}$ )  $\delta$  7.75–7.78 (m, 2H), 6.95–7.01 (m, 3H), 6.44–6.50 (m, 1H), 6.32–6.36 (m, 1H), 4.19 (dd,  $J = 1.6, 6.8$  Hz, 2H), 4.00 (dd,  $J = 1.6, 6.8$  Hz, 2H), 1.47 (s, 6H).  $m/z = 426.1$  ( $M + 1$ ).

**Synthesis of 10** ( $R_1=R_2=Me$ ). To a solution of 2-bromo-1-(4-fluorophenyl)ethanone (46.5 g, 214.29 mmol) in DMF (400 mL) was

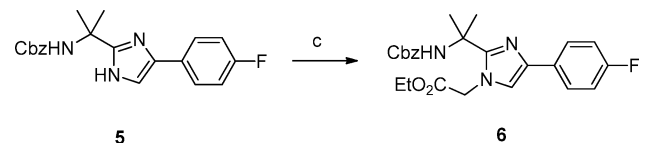


added 2-(benzyloxycarbonyl)-2-methylpropanoic acid (55.8 g, 236.4 mmol) and potassium carbonate (35.4 g, 256.5 mmol). The resulting solution was stirred for 4 h at rt. The resulting solution was diluted with 1000 mL of water. The resulting solution was extracted with ethyl acetate ( $2 \times 800$  mL), and the combined organic layer was washed with water ( $2 \times 800$  mL) and then brine ( $1 \times 800$  mL). The resulting



organic phase was concentrated under vacuum. This resulted in 67 g (84%) of compound **4** as a white solid.  $m/z = 374$  ( $M + 1$ ).

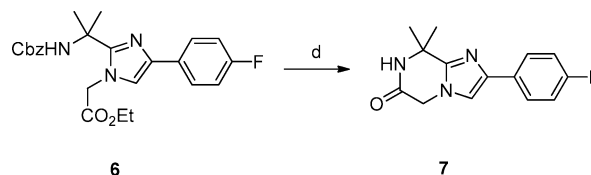
To a solution of compound **4** (70 g, 187.7 mmol) in toluene (700 mL) was added  $\text{NH}_4\text{OAc}$  (144.5 g, 1.88 mol). The resulting solution was heated to reflux for 3 h in an oil bath. The resulting mixture was concentrated under vacuum. The residue was dissolved in 800 mL of water. The resulting solution was extracted with ethyl acetate ( $2 \times 500$  mL), and the combined organic layer was washed with water ( $2 \times 800$  mL) and brine ( $1 \times 800$  mL). The resulting mixture was concentrated under vacuum. This resulted in 58 g (88%) of compound **5** as a white solid.  $^1\text{H}$  NMR



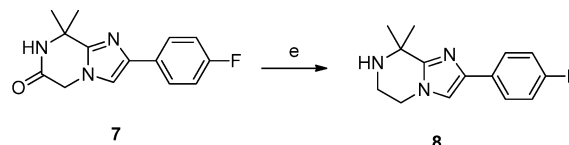
(300 Hz,  $\text{DMSO}-d_6$ )  $\delta$  11.91 (s, NH), 7.80–7.75 (m, 2H), 7.54–7.12 (m, 8H), 4.97 (m, 2H), 1.6 (s, 6H).

To a solution of compound **5** (58 g, 164.3 mmol) in DMF (400 mL) was added cesium carbonate (134 g, 411.0 mmol). This was followed by the addition of ethyl 2-bromoacetate (33 g, 197.6 mmol) dropwise with stirring at rt in 30 min. The resulting solution was stirred for 2 h at rt. The resulting solution was diluted with 1000 mL of water/ice. The resulting solution was extracted with ethyl acetate ( $2 \times 700$  mL), and the combined organic layer was washed with water ( $2 \times 800$  mL) and brine ( $1 \times 800$  mL). The resulting mixture was dried concentrated under vacuum. This resulted in 60 g (83%) of compound **6** as a yellow solid.  $^1\text{H}$  NMR (300 Hz,  $\text{DMSO}-d_6$ )  $\delta$  7.76–7.72 (m, 2H), 7.32–7.28 (m, 5H), 7.08–7.03 (m, 3H), 5.08 (m, 2H), 4.87 (m, 2H), 4.23 (q,  $J = 5.4$  Hz, 2H), 2.0 (s, 6H), 1.28 (t,  $J = 5.4$  Hz, 3H).

To a solution of compound **6** (70 g, 159.45 mmol, 1.00 equiv) in methanol (800 mL) was added palladium on carbon (10 g). The resulting solution was degassed and backfilled with hydrogen. The solution was stirred for 3 days at 25 °C. The solids were filtered out

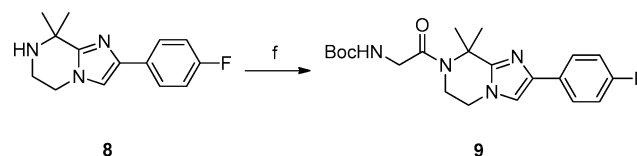


and washed with MeOH. The filtrate was concentrated under reduced pressure. This resulted in **7** (38 g, 145.56 mmol, 91%) as a white solid. LC-MS: (ES,  $m/z$ ) [ $M + H$ ] $^+$  calcd for  $\text{C}_{14}\text{H}_{14}\text{FN}_3\text{O}$ , 260; found, 260.



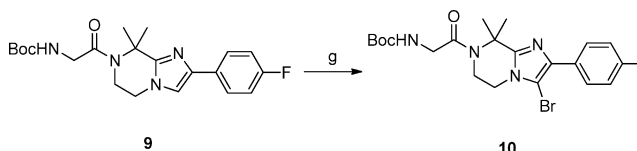
$^1\text{H}$  NMR (300 Hz,  $\text{DMSO}-d_6$ )  $\delta$  7.79–7.74 (m, 2H), 7.13–7.07 (m, 3H), 6.35 (s, 1H), 4.73 (s, 2H), 1.79 (s, 6H).

To a stirred solution of compound **7** (5 g, 19.2 mmol, 1 equiv) was dissolved in 50 mL of THF, 1 M borane/THF complex (57 mL, 57 mmol, 3 equiv) was added slowly and reaction was refluxed overnight. LCMS indicated that the reaction was complete. THF was removed under reduced pressure. The reaction was quenched with MeOH. The crude product of **8** (4.5 g, 18.3 mmol, 95%) was used in the next step. LC-MS: (ES,  $m/z$ ) [ $M + H$ ] $^+$  calcd for  $\text{C}_{14}\text{H}_{17}\text{FN}_3$ , 246; found, 246.  $^1\text{H}$  NMR ( $\text{DMSO}$ , 300 Hz)  $\delta$  7.75–7.70 (m, 2H), 7.4 (s, 1H), 7.14 (t,  $J =$



9.0 Hz, 2H), 3.9 (t,  $J = 5.4$  Hz, 2H), 2.51 (t,  $J = 5.4$  Hz, 2H), 1.41 (s, 6H); NH proton not observed.

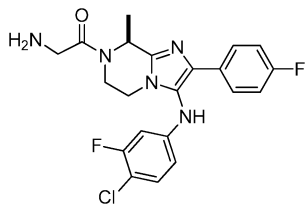
To a stirred solution of compound **8** (2.9 g, 11.82 mmol, 1.1 equiv) and 2-(*tert*-butoxycarbonylamino)acetic acid (2.27 g, 13 mmol, 1.1 equiv) in 15 mL of dichloromethane were added DIEA (2.47 mL, 14.18 mmol, 1.2 equiv) and HATU (5.39 g, 14.18 mmol, 1.2 equiv). The reaction mixture was stirred at rt for 8 h. HPLC/MS analysis showed that desired product compound **9** was the major product. The reaction was further diluted with dichloromethane (70 mL). The organic layer was washed with water ( $1 \times 30$  mL), followed by saturated  $\text{NaHCO}_3$  ( $1 \times 30$  mL) and finally with brine ( $1 \times 30$  mL). The organic layer was then dried over anhydrous  $\text{Na}_2\text{SO}_4$  and concentrated in vacuo. The resulting oil was purified using column chromatography with hexanes/ethyl acetate (0–100% linear gradient) used as eluant. The desired product compound **9** was obtained as oil (3.3 g, 8.27 mmol, 70%).  $^1\text{H}$  NMR (300 Hz,  $\text{DMSO}-d_6$ )  $\delta$  7.77–7.72 (m, 2H), 7.54 (s, 1H), 7.20–7.14 (m, 2H), 6.84–6.80 (m, 1H), 4.07 (s, 2H), 3.90 (d,  $J = 3.0$  Hz, 2H), 3.70 (s, 2H), 1.80 (s, 6H), 1.40 (s,



9H). LC-MS: (ES,  $m/z$ ) [ $M + H$ ] $^+$  calcd for  $\text{C}_{21}\text{H}_{28}\text{FN}_4\text{O}_3$ , 403; found, 403.

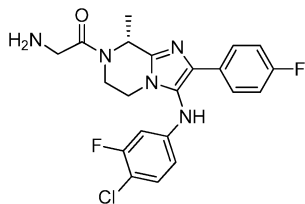
To a stirred solution of compound **9** (3.02 g, 7.51 mmol) in dichloromethane (30 mL) was added  $\text{Br}_2$  (0.43 mL, 8.26 mmol) in acetic acid (3 mL). The reaction mixture was stirred at rt for 30 min.

HPLC/MS test showed that desired product (II) was the only peak. Solvent was removed via rotavap at a temperature no higher than 20 °C. After neutralization, white solid (3.60 g) was obtained. The product was confirmed by 400 MHz proton NMR to be the title compound **10**. The product was used in the next step without further purification and exhibited quantitative mass recovery. LC-MS: (ES,  $m/z$ )  $[M + H]^+$  calcd for  $C_{21}H_{27}BrFN_4O_3$ , 482; found, 482.  $^1H$  NMR (MeOH- $d_4$ , 400 Hz)  $\delta$  7.84–7.81 (m, 2H), 7.14 (t,  $J = 8.8$  Hz, 2H), 4.09–4.01 (m, 4H), 3.81 (t,  $J = 4.8$  Hz, 2H), 1.89 (s, 6H), 1.46 (s, 9H).



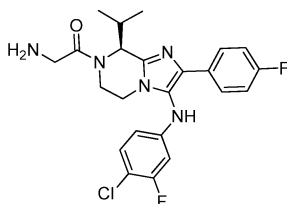
**Synthesis of 17.** (S)-2-Amino-1-(3-(4-chloro-3-fluorophenylamino)-2-(4-fluorophenyl)-8-methyl-5,6-dihydroimidazo[1,2-a]pyrazin-7(8H)-yl)ethanone:

Compound **17** was prepared from compound **10** ( $R_1 = Me$ ,  $R_2 = H$ ) by a  $Pd_2(dba)_3$  mediated amination reaction with 4-chloro-3-fluoroaniline followed by a TFA mediated deprotection in a protocol similar to compound **22**.  $^1H$  NMR (MeOH- $d_4$ , 400 MHz)  $\delta$  7.61–7.58 (m, 2H), 7.16–7.12 (m, 1H), 7.07–7.02 (m, 2H), 6.47–6.42 (m, 2H), 5.74 (m, 1H), 4.07–3.68 (m, 6H), 1.56 (d,  $J = 6.8$  Hz, 3H).  $m/z = 432.0$  ( $M + 1$ ).



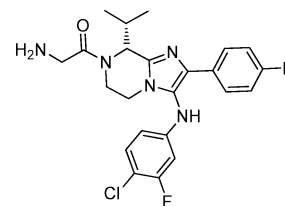
**Synthesis of 18.** (R)-2-Amino-1-(3-(4-chloro-3-fluorophenylamino)-2-(4-fluorophenyl)-8-methyl-5,6-dihydroimidazo[1,2-a]pyrazin-7(8H)-yl)ethanone:

Compound **18** was prepared from compound **10** ( $R_1 = Me$ ,  $R_2 = H$ ) by a  $Pd_2(dba)_3$  mediated amination reaction with 4-chloro-3-fluoroaniline followed by a TFA mediated deprotection in a protocol similar to compound **22**.  $m/z = 432.2$  ( $M + 1$ ).



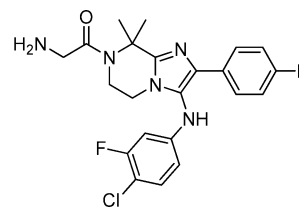
**Synthesis of 19.** (S)-2-Amino-1-(3-(4-chloro-3-fluorophenylamino)-2-(4-fluorophenyl)-8-isopropyl-5,6-dihydroimidazo[1,2-a]pyrazin-7(8H)-yl)ethanone:

Compound **19** was prepared from compound **10** ( $R_1 = isopropyl$ ,  $R_2 = H$ ) by a  $Pd_2(dba)_3$  mediated amination reaction with 4-chloro-3-fluoroaniline followed by a TFA mediated deprotection in a protocol similar to compound **22**.  $^1H$  NMR (MeOH- $d_4$ , 400 MHz)  $\delta$  7.61–7.58 (m, 2H), 7.26–7.20 (m, 3H), 6.45–6.38 (m, 2H), 5.5 (d,  $J = 7.9$  Hz, 1H), 4.11–3.75 (m, 6H), 2.36–2.31 (m, 1H), 1.15 (d,  $J = 6.7$  Hz, 3H), 0.95 (d,  $J = 6.7$  Hz, 3H).  $m/z = 460.2$  ( $M + 1$ ).



**Synthesis of 20.** (R)-2-Amino-1-(3-(4-chloro-3-fluorophenylamino)-2-(4-fluorophenyl)-8-isopropyl-5,6-dihydroimidazo[1,2-a]pyrazin-7(8H)-yl)ethanone:

Compound **20** was prepared from compound **10** ( $R_1 = isopropyl$ ,  $R_2 = H$ ) by a  $Pd_2(dba)_3$  mediated amination reaction with 4-chloro-3-fluoroaniline followed by a TFA mediated deprotection in a protocol similar to compound **22**.  $^1H$  NMR (MeOH- $d_4$ , 400 MHz)  $\delta$  7.69–7.67 (m, 2H), 7.27–7.20 (m, 3H), 6.68–6.65 (dd,  $J = 2.5$  Hz,  $J = 11.0$  Hz, 1H), 6.60–6.55 (dd,  $J = 2.25$  Hz,  $J = 8.7$  Hz, 1H), 5.78 (d,  $J = 7.7$  Hz, 1H), 4.42–4.02 (m, 4H), 3.94–3.87 (m, 2H), 2.55 (m, 1H), 1.24 (d,  $J = 6.8$  Hz, 3H), 1.05 (d,  $J = 6.8$  Hz, 3H).  $m/z = 460.2$  ( $M + 1$ ).

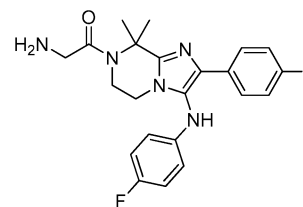


**Synthesis of 21.** 2-Amino-1-(3-(4-chloro-3-fluorophenylamino)-2-(4-fluorophenyl)-8,8-dimethyl-5,6-dihydroimidazo[1,2-a]pyrazin-7(8H)-yl)ethanone:

Compound **21** was prepared from compound **10** ( $R_1 = Me$ ,  $R_2 = Me$ ) by a  $Pd_2(dba)_3$  mediated amination reaction with 4-chloro-3-fluoroaniline followed by a TFA mediated deprotection in a protocol similar to compound **22**.

$^1H$  NMR (MeOH- $d_4$ , 400 MHz)  $\delta$  7.59–7.55 (m, 2H), 7.20–7.13 (m, 3H), 6.60 (dd,  $J = 11.2$ , 10.8 Hz, 1H), 6.53 (dd,  $J = 2.4$ , 0.8 Hz, 1H), 4.0 (m, 4H), 3.78 (m, 2H), 2.00 (s, 6H).  $m/z = 446.0$  ( $M + 1$ ). Elemental analysis (compound + 2.0 HCl + 2.0 H<sub>2</sub>O): % C, 47.62; % H, 5.09; %N, 12.62; (calc). %C = 47.54/47.38; %N = 12.44/12.46; %H = 4.62/4.57 (experimental). LC/MS major mass 446.2.

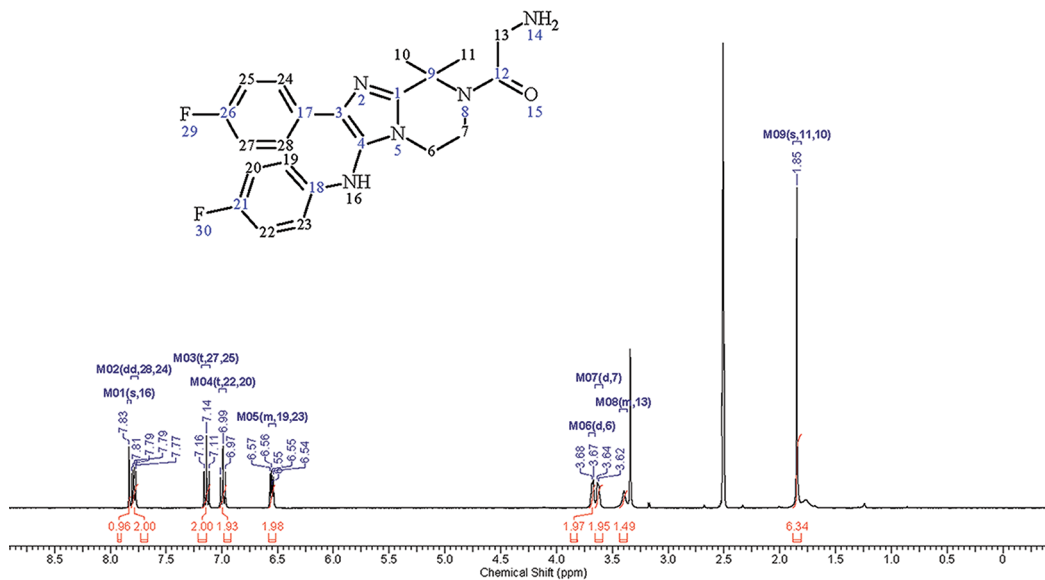
**Synthesis of 22 (Analogue of Compound 12 in Scheme 2 where  $R_1 = R_2 = Me$ ).** 2-Amino-1-(2-(4-fluorophenyl)-3-(4-fluorophenylamino)-8,8-dimethyl-5,6-dihydroimidazo[1,2-a]pyrazin-7(8H)-yl)ethanone:



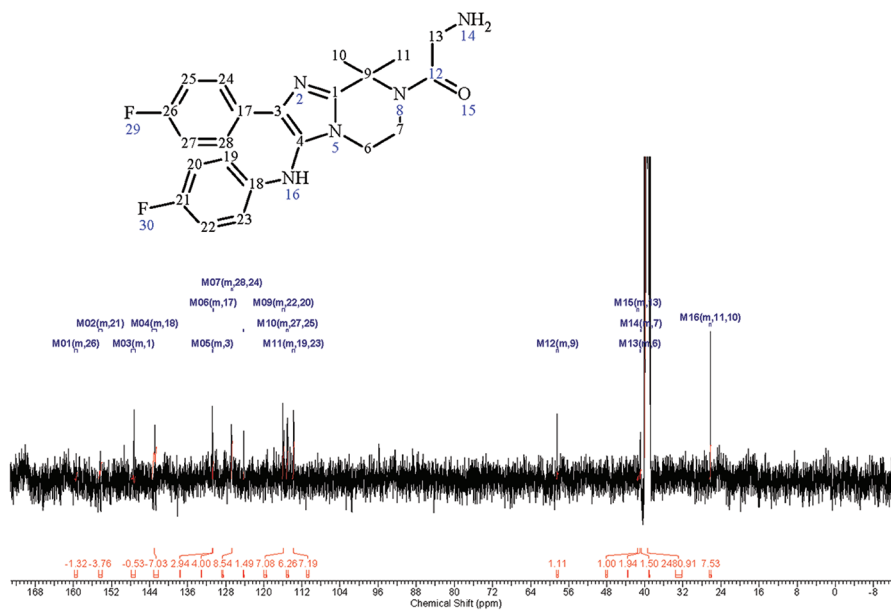
In a glass vial,  $Cs_2CO_3$ , 4F-aniline (0.462 g, 4.1 mmol, 2.0 equiv),  $Pd_2(dba)_3$  (0.095 g, 0.104 mmol, 0.05 equiv), xantphos (0.120 g, 0.208 mmol, 0.1 equiv), and dioxane were stirred for 5 min at rt. Compound **10** (1 g, 2.08 mmol, 1.0 equiv) was added to the reaction mixture, after which the reaction mixture was degassed for 15 min and then stirred at 120 °C under  $N_2$  for 8 h. HPLC/MS test showed that the starting material compound **10** was consumed and desired product was formed predominantly along with some **9**. The reaction was filtered to remove solids. The reaction mixture was concentrated and then purified by normal phase column chromatography (silica gel 80 g) using a gradient of 100–0% to 0–100% hexane:EtOAc. The desired product eluted at 60:40 EtOAc:hexanes. The organic layer was concentrated at reduced pressure to yield the Boc derivative (950 mg, 89%) yield.

Table 8. Regiochemistry of the diMe Groups Assigned Using 2-D NMR Techniques

$^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ )  $\delta$  ppm 1.85 (s, 16 H) 3.37 - 3.43 (m, 4 H) 3.63 (d,  $J=5.12$  Hz, 5 H) 3.68 (d,  $J=5.12$  Hz, 5 H) 6.32 - 6.58 (m, 5 H) 6.99 (t,  $J=8.88$  Hz, 5 H) 7.14 (t,  $J=8.96$  Hz, 5 H) 7.79 (dd,  $J=8.96, 5.72$  Hz, 5 H) 7.83 (s, 2 H)



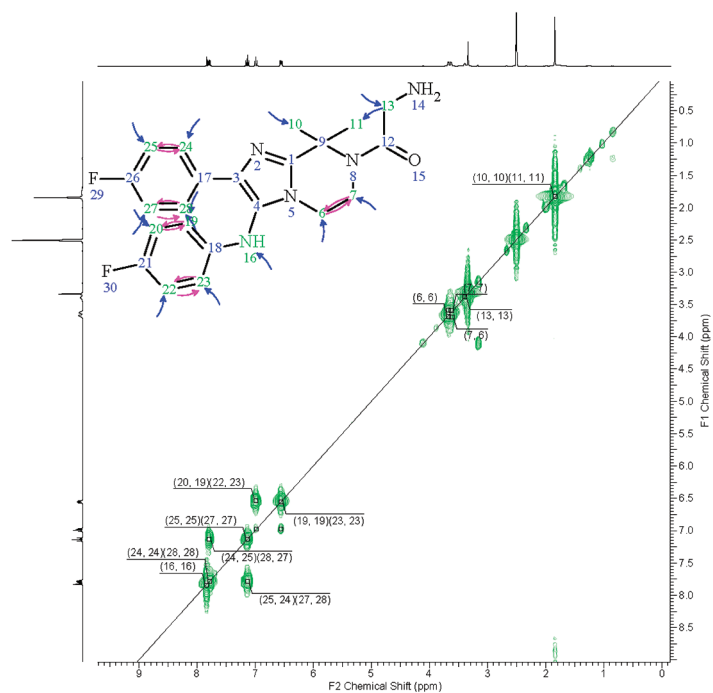
No.	Atom	Exp. Shift (ppm)	Multiplet	No.	Atom	Exp. Shift (ppm)	Multiplet
1	6	3.68	M06	8	20	6.99	M04
2	7	3.63	M07	9	22	6.99	M04
3	10	1.85	M09	10	23	6.55	M05
4	11	1.85	M09	11	24	7.79	M02
5	13	3.4	M08	12	25	7.14	M03
6	16	7.83	M01	13	27	7.14	M03
7	19	6.55	M05	14	28	7.79	M02



No.	Atom	Exp. Shift (ppm)	Multiplet	No.	Atom	Exp. Shift (ppm)	Multiplet
1	1	147.43	M03	3	4	124.26	M08
2	3	130.8	M05	4	6	40.99	M13

Table 8. continued

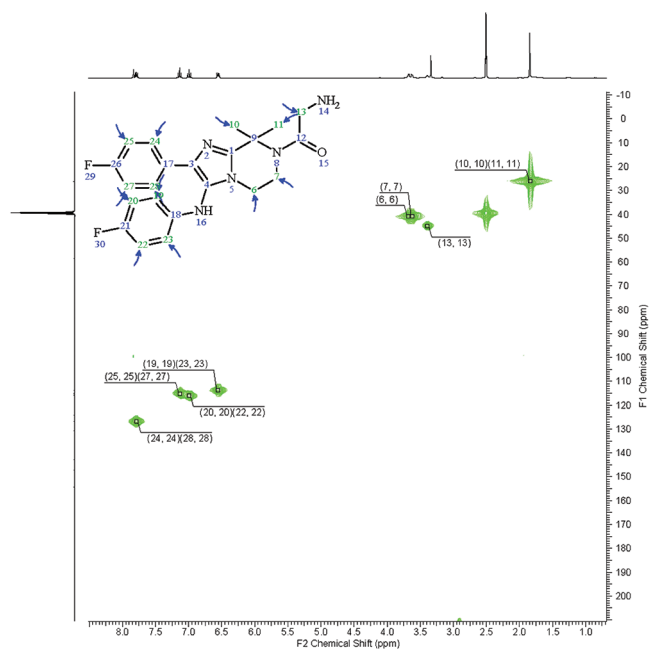
No.	Atom	Shift (ppm)	Multiplet	No.	Atom	Shift (ppm)	Multiplet
5	7	40.83	M14	14	21	154.35	M02
6	9	58.4	M12	15	22	115.9	M09
7	10	26.24	M16	16	23	113.76	M11
8	11	26.24	M16	17	24	126.71	M07
9	13	41.48	M15	18	25	115.04	M10
10	17	130.64	M06	19	26	159.44	M01
11	18	143.03	M04	20	27	115.04	M10
12	19	113.76	M11	21	28	126.71	M07
13	20	115.9	M09				



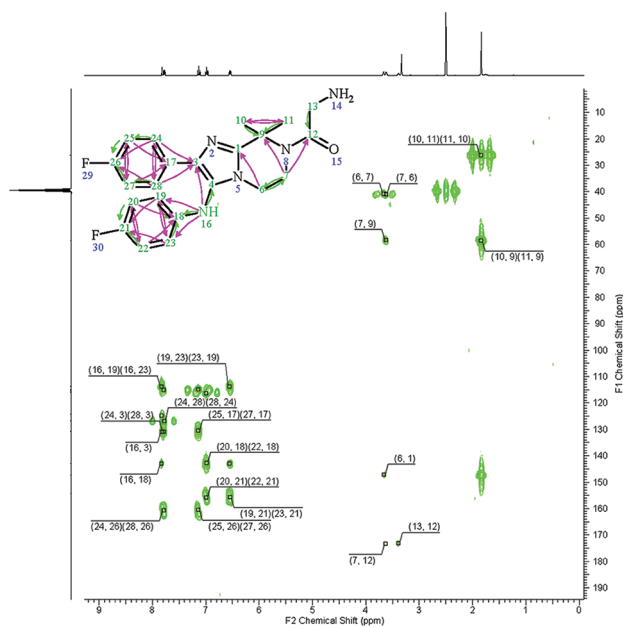
No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	6	6	3.67	3.68	13	22	22	6.98	6.98
2	7	6	3.62	3.69	14	23	22	6.55	6.98
3	6	7	3.68	3.59	15	22	23	6.99	6.54
4	7	7	3.62	3.59	16	23	23	6.55	6.56
5	10	10	1.84	1.82	17	24	24	7.78	7.78
6	11	11	1.84	1.82	18	25	24	7.14	7.78
7	13	13	3.39	3.38	19	24	25	7.79	7.13
8	16	16	7.83	7.84	20	25	25	7.14	7.13
9	19	19	6.55	6.56	21	27	27	7.14	7.13
10	20	19	6.99	6.54	22	28	27	7.79	7.13
11	19	20	6.55	6.98	23	27	28	7.14	7.78
12	20	20	6.98	6.98	24	28	28	7.78	7.78
13	22	22	6.98	6.98					



Table 8. continued



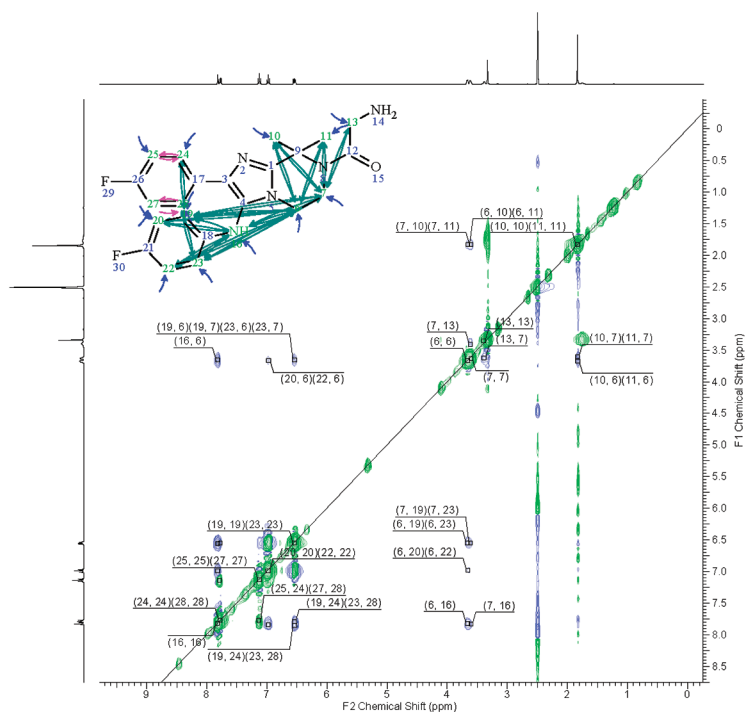
No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	6	6	3.68	40.89	8	22	22	6.99	115.94
2	7	7	3.62	40.92	9	23	23	6.55	113.69
3	10	10	1.84	26.04	10	24	24	7.79	126.84
4	11	11	1.84	26.04	11	25	25	7.13	115.12
5	13	13	3.39	44.88	12	27	27	7.13	115.12
6	19	19	6.55	113.69	13	28	28	7.79	126.84
7	20	20	6.99	115.94					



No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	6	1	3.66	147.14	3	24	3	7.79	130.83
2	16	3	7.82	130.82	4	28	3	7.79	130.83

Table 8. continued

No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
5	16	4	7.83	124.77	24	22	20	6.99	116.22
6	7	6	3.62	40.96	25	19	21	6.55	155.68
7	6	7	3.67	40.7	26	20	21	6.99	155.86
8	7	9	3.62	58.38	27	22	21	6.99	155.86
9	10	9	1.84	58.5	28	23	21	6.55	155.68
10	11	9	1.84	58.5	29	20	22	6.99	116.22
11	11	10	1.84	26.29	30	16	23	7.83	113.74
12	10	11	1.84	26.29	31	19	23	6.56	113.75
13	7	12	3.63	173.36	32	28	24	7.78	126.82
14	13	12	3.39	173.07	33	24	25	7.79	114.99
15	25	17	7.14	130.47	34	27	25	7.14	114.94
16	27	17	7.14	130.47	35	24	26	7.79	160.61
17	16	18	7.83	142.87	36	25	26	7.14	160.51
18	19	18	6.55	143.02	37	27	26	7.14	160.51
19	20	18	6.99	142.8	38	28	26	7.79	160.61
20	22	18	6.99	142.8	39	25	27	7.14	114.94
21	23	18	6.55	143.02	40	28	27	7.79	114.99
22	16	19	7.83	113.74	41	24	28	7.78	126.82
23	23	19	6.56	113.75					



No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	6	6	3.66	3.66	3	11	6	1.83	3.67
2	10	6	1.83	3.67	4	16	6	7.82	3.66

Table 8. continued

No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
5	19	6	6.54	3.65	31	19	19	6.54	6.55
6	20	6	6.97	3.66	32	24	19	7.78	6.54
7	22	6	6.97	3.66	33	6	20	3.66	6.98
8	23	6	6.54	3.65	34	16	20	7.82	6.99
9	7	7	3.61	3.63	35	20	20	6.98	7
10	10	7	1.83	3.6	36	6	22	3.66	6.98
11	11	7	1.83	3.6	37	16	22	7.82	6.99
12	13	7	3.39	3.63	38	22	22	6.98	7
13	19	7	6.54	3.65	39	6	23	3.66	6.55
14	23	7	6.54	3.65	40	7	23	3.61	6.54
15	6	10	3.66	1.83	41	16	23	7.83	6.56
16	7	10	3.61	1.83	42	23	23	6.54	6.55
17	10	10	1.84	1.84	43	28	23	7.78	6.54
18	6	11	3.66	1.83	44	19	24	6.55	7.79
19	7	11	3.61	1.83	45	19	24	6.54	7.85
20	11	11	1.84	1.84	46	24	24	7.78	7.77
21	7	13	3.61	3.41	47	25	24	7.14	7.77
22	13	13	3.4	3.35	48	24	25	7.79	7.14
23	6	16	3.66	7.83	49	25	25	7.13	7.13
24	7	16	3.61	7.83	50	27	27	7.13	7.13
25	16	16	7.82	7.83	51	28	27	7.79	7.14
26	20	16	6.98	7.84	52	23	28	6.55	7.79
27	22	16	6.98	7.84	53	23	28	6.54	7.85
28	6	19	3.66	6.55	54	27	28	7.14	7.77
29	7	19	3.61	6.54	55	28	28	7.78	7.77
30	16	19	7.83	6.56					

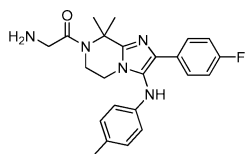
LC-MS: (ES,  $m/z$ )  $[M + H]^+$  calcd for  $C_{27}H_{32}F_2N_5O_3$ , 512; found, 512.

The Boc compound was treated with 20% TFA in  $CH_2Cl_2$  (50 mL) and was added to the mixture. After the completion of this reaction (monitored by LCMS), the resulting mixture was concentrated under reduced pressure. The resulting residue was purified by reverse phase HPLC to yield product as a TFA salt. The acetonitrile–water layer was concentrated in vacuo to remove all the solvents. The residue

was dissolved in dichloromethane and carefully neutralized by satd  $NaHCO_3$ . The organic layer was successively washed with brine followed by water. The organic layer was concentrated to yield **22** (450 mg, 52%). LC-MS: (ES,  $m/z$ )  $[M + H]^+$  calcd for  $C_{22}H_{23}F_2N_5O$ , 412.2; found, 412.1.  $^1H$  NMR (MeOH- $d_4$ , 400 Hz)  $\delta$  7.61–7.57 (m, 2H), 6.94 (t,  $J = 8.8$  Hz, 2H), 6.81 (t,  $J = 8.8$  Hz, 2H), 6.47 (m, 2H), 3.72 (m, 2H), 3.58 (m, 2H), 3.42 (m, 2H), 1.85 (s, 6H). Elemental analysis compound **22** with 0.65 equiv  $H_2O$ : C, 62.44; N, 16.55; H,

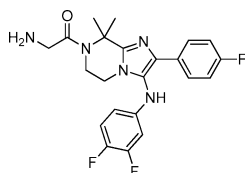
5.79 (calculated). C = 62.54/62.44; N = 16.35/16.29; H = 5.52/5.61 (experimental).

The regiochemistry of the diMe groups was unambiguously assigned using 2-D NMR techniques (Table 8).



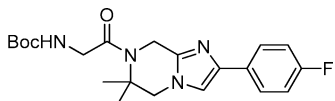
**Synthesis of 23.** 2-Amino-1-(2-(4-fluorophenyl)-8,8-dimethyl-3-(*p*-tolylamino)-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8*H*)-yl)-ethanone:

Compound **23** was prepared from compound **10** ( $R_1 = \text{Me}$ ,  $R_2 = \text{Me}$ ) by a  $\text{Pd}_2(\text{dba})_3$  mediated amination reaction with 4-methylaniline followed by a TFA mediated deprotection in a protocol similar to compound **22**.  $^1\text{H NMR}$  ( $\text{MeOH}-d_4$ , 400 MHz)  $\delta$  7.79–7.69 (m, 2H), 7.21 (t,  $J = 8.4$  Hz, 2H), 7.04 (d,  $J = 8.0$  Hz, 2H), 6.76 (d,  $J = 8.2$  Hz, 2H), 4.11 (s, 2H), 3.88 (m, 4H), 2.22 (s, 3H), 2.12 (s, 6H). LC/MS major mass: 408.1 ( $M + \text{H}$ ).



**Synthesis of 24.** 2-Amino-1-(3-((3,4-difluorophenyl)amino)-2-(4-fluorophenyl)-8,8-dimethyl-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8*H*)-yl)ethanone:

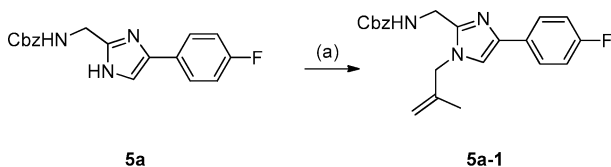
Compound **24** was prepared from compound **10** ( $R_1 = \text{Me}$ ,  $R_2 = \text{Me}$ ) by a  $\text{Pd}_2(\text{dba})_3$  mediated amination reaction with 3,4-difluoroaniline followed by a TFA mediated deprotection in a protocol similar to compound **22**.  $^1\text{H NMR}$  ( $\text{MeOH}-d_4$ , 400 MHz)  $\delta$  7.73–7.69 (m, 2H), 7.21 (t,  $J = 8.8$  Hz, 2H), 7.06–7.03 (m, 1H), 6.76–6.65 (m, 1H), 6.53–6.51 (m, 1H), 4.08–4.06 (m, 4H), 3.86–3.83 (m, 2H), 2.07 (s, 6H). Elemental analysis (compound + 2.9 HCl): %C, 49.37; %H, 4.69; %N, 13.09; (calc). %C = 49.43/49.06; %N = 12.89/12.77; %H = 4.7/4.81 (experimental).  $m/z = 430.2$  ( $M + 1$ ).



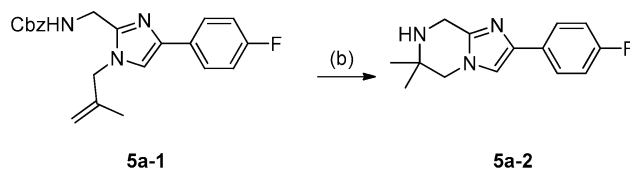
**Synthesis of 13.** *tert*-Butyl 2-(3-bromo-2-(4-fluorophenyl)-6,6-dimethyl-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8*H*)-yl)-2-oxoethylcarbamate.

**13** was prepared from compound **5a** by the following way:

To a stirred solution of compound **5a** (1.1 g, 3.38 mmol, 1.00 equiv) in DMF (30 mL) was added 3-chloro-2-methylprop-1-ene (500 mg, 5.49 mmol, 1.50 equiv), potassium carbonate (560 mg, 4.06 mmol, 1.10 equiv), and potassium iodide (1.12 g, 6.75 mmol, 2.00 equiv) at rt. The reaction mixture was stirred for 48 h at 40 °C. The reaction mixture was diluted with ethyl acetate (100 mL). The mixture was washed with brine ( $3 \times 10$  mL), dried over

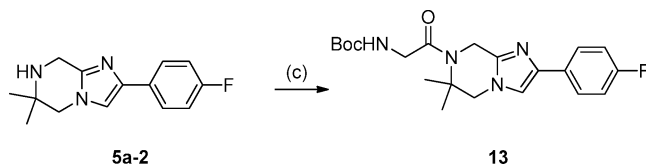


sodium sulfate, and concentrated under vacuum. The residue was applied onto a silica gel column with petroleum ether/EtOAc



(5:1) to give 0.7 g (55%) of compound **5a-1** as a light-yellow solid.  $m/z = 380$  ( $M + 1$ ).

Into a 30 mL sealed tube, was placed compound **5a-1** (2.0 g, 5.28 mmol, 1.00 equiv), acetic acid (12 mL), and methanesulfonic acid (2 mL). The reaction mixture was stirred for 12 h at 260 °C (the temperature of the sand bath). The reaction mixture was cooled to rt. The mixture was poured into 20 mL of water. The aqueous layer was washed with ethyl acetate ( $3 \times 10$  mL). Aqueous sodium hydroxide (1N) was added to adjust pH to 8. The aqueous layer was extracted with ethyl acetate ( $3 \times 10$  mL). The combined organic layers were washed with brine ( $3 \times 10$  mL), dried over sodium sulfate, and concentrated under vacuum. The solid was collected by filtration and washed with 5 mL of *n*-hexane to give 500 mg (39%) of compound **5a-2** as a white solid.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.70–7.65 (m, 2H),



7.04–6.9 (m, 2H), 6.96 (s, 1H), 4.12 (s, 2H), 3.69 (s, 2H), 1.23 (s, 6H).

To a stirred solution of compound **5a-2** (280 mg, 1.14 mmol, 1.0 equiv) in DMF (10 mL) was added 2-(*tert*-butoxycarbonyl)-acetic acid (600 mg, 3.43 mmol, 3.0 equiv), HATU (1.3 g, 3.42 mmol, 3.0 equiv), and DIEA (880 mg, 6.82 mmol, 6.0 equiv) at rt. The reaction mixture was stirred overnight at rt. The reaction mixture was diluted with ethyl acetate (100 mL). The organic layer was washed with brine ( $3 \times 10$  mL), dried over sodium sulfate, and concentrated under vacuum. The residue was applied onto a silica gel column with  $\text{CH}_2\text{Cl}_2/\text{MeOH}$  (10:1) to give 280 mg (57%) of compound **13** as a brown solid.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.72–7.68 (m, 2H), 7.23 (s, 1H), 7.13–7.07 (m, 2H), 5.46 (s, NH), 4.65 (s, 2H), 4.05–4.04 (m, 2H), 3.97 (s, 2H), 1.55–1.39 (m, 15H).

The structure of **13** was verified by HMQC, COSY, HMBC, and ROESY (Table 9).

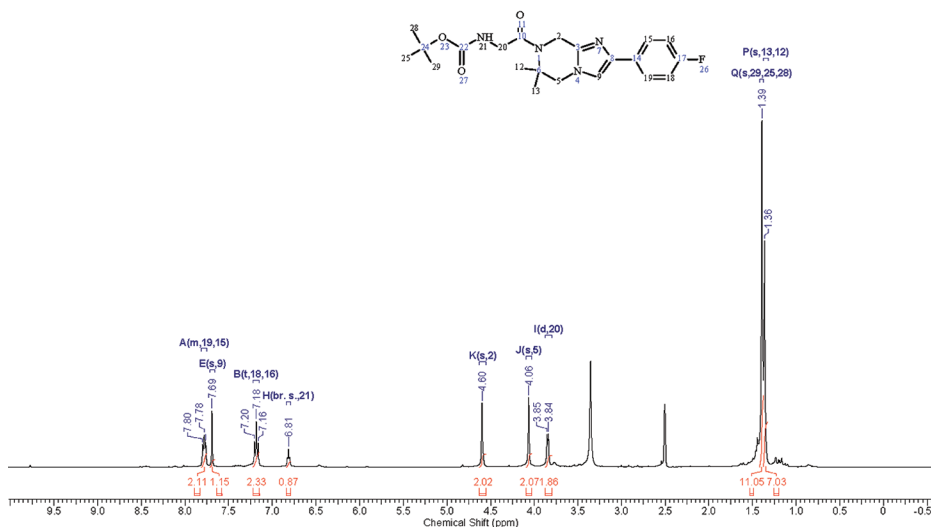
**Synthesis of 25 (Analogue of Compound 14 in Scheme 3).** 2-Amino-1-(2-(4-fluorophenyl)-3-(4-fluorophenylamino)-6,6-dimethyl-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8*H*)-yl)-ethanone:

To a stirred solution of compound **13** (386 g, 0.96 mmol) in 6 mL of dichloromethane was added  $\text{Br}_2$  (55  $\mu\text{L}$ , 1.06 mmol) in acetic acid (2 mL). The reaction mixture was stirred at rt for 30 min. Solvent was removed via rotavap at a temperature no higher than 20 °C. After neutralization, the residue was subjected to flash chromatography (40 g, 0–100% ethyl acetate in hexane, 50 min, dry loading) purification to give 256 mg (55%) of the title compound **13a** as a colorless oil.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.87–7.83 (m, 2H), 7.03 (t,  $J = 8.8$  Hz, 2H), 5.46 (s, 1H), 4.5 (s, 2H), 3.97 (d,  $J = 4.2$  Hz, 2H), 3.81 (s, 2H), 1.45 (s, 6H), 1.39 (s, 9H).

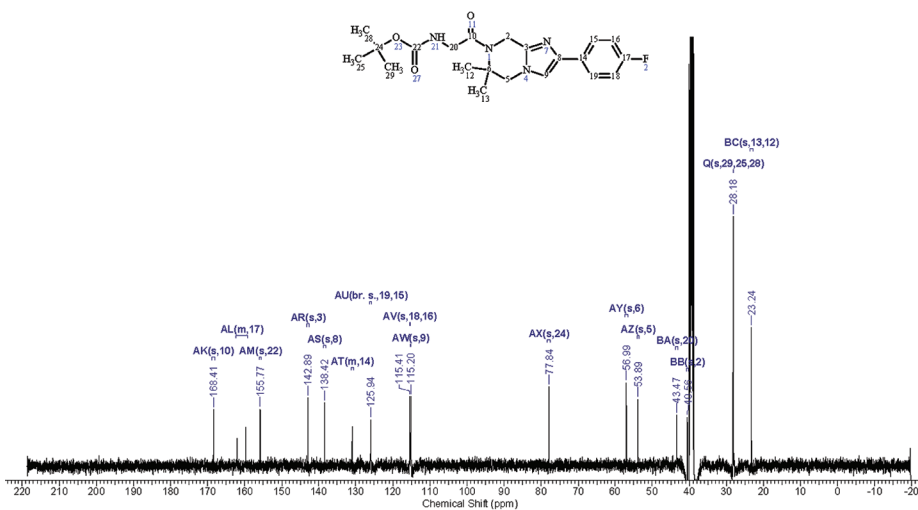


Table 9. Structure of 13 Verified by HMQC, COSY, HMBC, and ROESY

<sup>1</sup>H NMR (400 MHz, DMSO-*d*<sub>6</sub>) δ ppm 1.36 (s, 6 H) 1.39 (s, 10 H) 3.85 (d, *J*=5.62 Hz, 2 H) 4.06 (s, 2 H) 4.60 (s, 2 H) 6.81 (br. s., 1 H) 7.18 (t, *J*=8.68 Hz, 2 H) 7.69 (s, 1 H) 7.75 - 7.81 (m, 2 H)



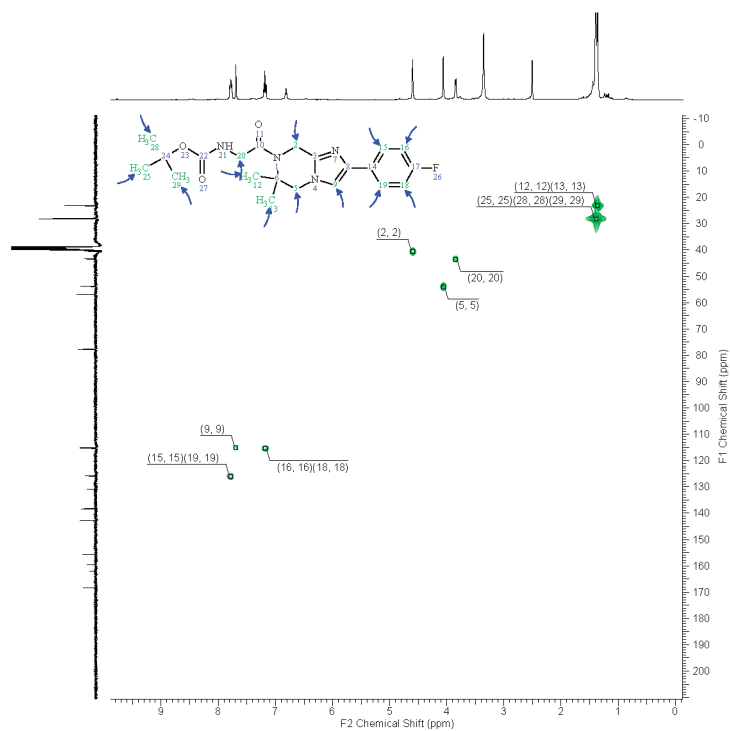
No.	Atom	Exp. Shift (ppm)	Multiplet	No.	Atom	Exp. Shift (ppm)	Multiplet
1	2	4.6	K	8	18	7.18	B
2	5	4.06	J	9	19	7.79	A
3	9	7.69	E	10	20	3.85	I
4	12	1.36	P	11	21	6.81	H
5	13	1.36	P	12	25	1.39	Q
6	15	7.79	A	13	28	1.39	Q
7	16	7.18	B	14	29	1.39	Q



No.	Atom	Exp. Shift (ppm)	Multiplet	No.	Atom	Exp. Shift (ppm)	Multiplet
1	2	40.56	BB	3	5	53.89	AZ
2	3	142.89	AR	4	6	56.99	AY

Table 9. continued

No.	Atom	Exp. Shift (ppm)	Multiplet	No.	Atom	Exp. Shift (ppm)	Multiplet
5	8	138.42	AS	14	18	115.41	AV
6	9	115.2	AW	15	19	125.94	AU
7	10	168.41	AK	16	20	43.47	BA
8	12	23.24	BC	17	22	155.77	AM
9	13	23.24	BC	18	24	77.84	AX
10	14	130.93	AT	19	25	28.18	Q
11	15	125.94	AU	20	28	28.18	Q
12	16	115.41	AV	21	29	28.18	Q
13	17	160.78	AL				



No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	2	2	4.6	40.56	8	18	18	7.18	115.38
2	5	5	4.06	53.91	9	19	19	7.79	125.97
3	9	9	7.69	115.16	10	20	20	3.85	43.5
4	12	12	1.36	23.16	11	25	25	1.39	28.18
5	13	13	1.36	23.16	12	28	28	1.39	28.18
6	15	15	7.79	125.97	13	29	29	1.39	28.18
7	16	16	7.18	115.38					

Table 9. continued

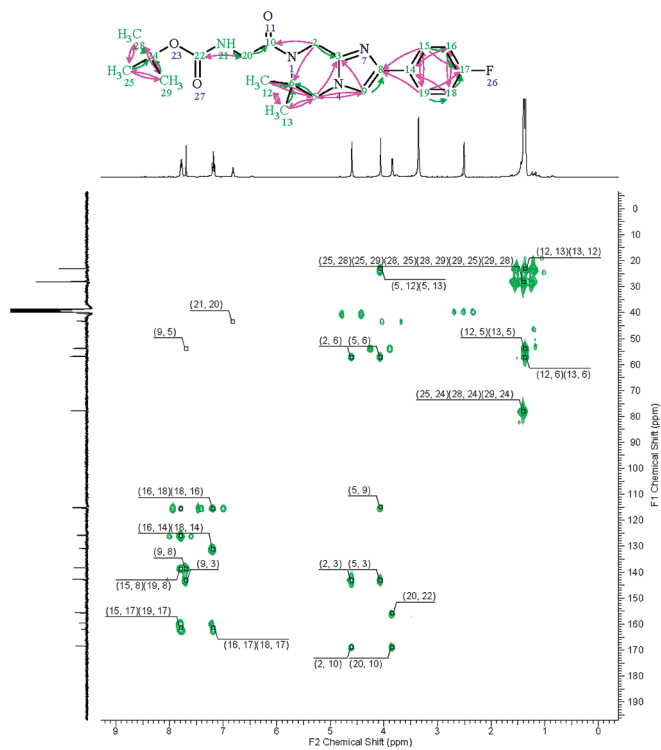
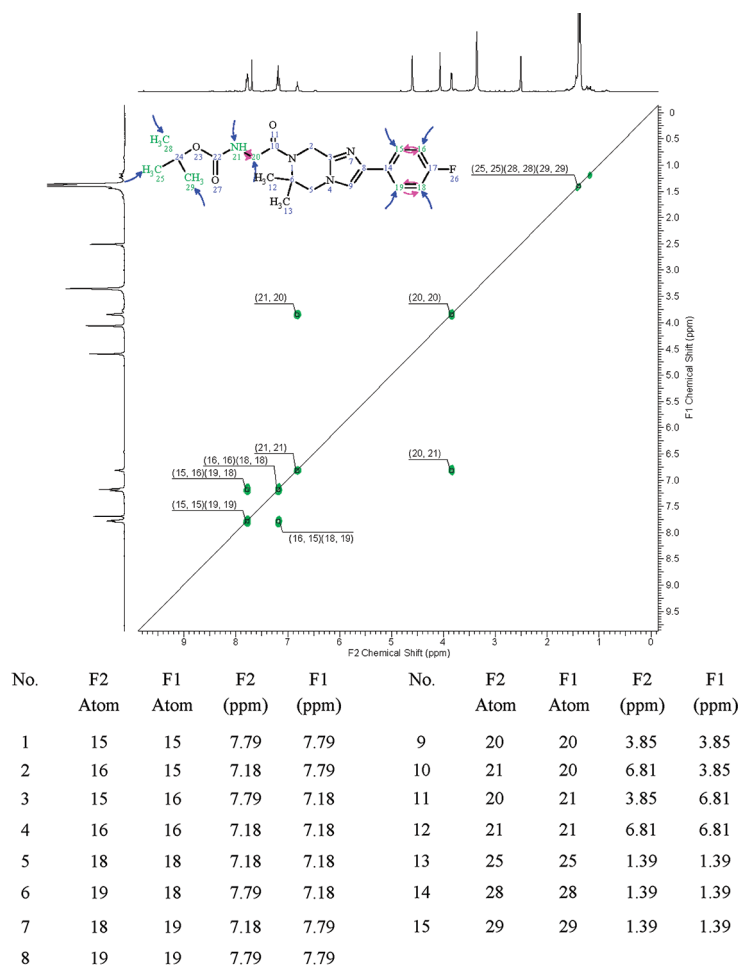


Table 9. continued

No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	2	3	4.6	143.21	22	18	14	7.18	131.16
2	5	3	4.06	143.21	23	19	15	7.79	125.97
3	9	3	7.69	143.21	24	15	16	7.79	115.38
4	9	5	7.69	53.91	25	18	16	7.18	115.38
5	12	5	1.36	53.91	26	15	17	7.79	161.42
6	13	5	1.36	53.91	27	16	17	7.18	161.42
7	2	6	4.6	57.29	28	18	17	7.18	161.42
8	5	6	4.06	57.29	29	19	17	7.79	161.42
9	12	6	1.36	57.29	30	16	18	7.18	115.38
10	13	6	1.36	57.29	31	19	18	7.79	115.38
11	9	8	7.69	138.7	33	21	20	6.81	43.5
12	15	8	7.79	138.7	34	20	22	3.85	155.84
13	19	8	7.79	138.7	35	25	24	1.39	77.91
14	5	9	4.06	114.89	36	28	24	1.39	77.91
15	2	10	4.6	168.77	37	29	24	1.39	77.91
16	20	10	3.85	168.77	38	28	25	1.39	28.18
17	5	12	4.06	23.16	39	29	25	1.39	28.18
18	13	12	1.36	23.16	40	25	28	1.39	28.18
19	5	13	4.06	23.16	41	29	28	1.39	28.18
20	12	13	1.36	23.16	42	25	29	1.39	28.18
21	16	14	7.18	131.16	43	28	29	1.39	28.18

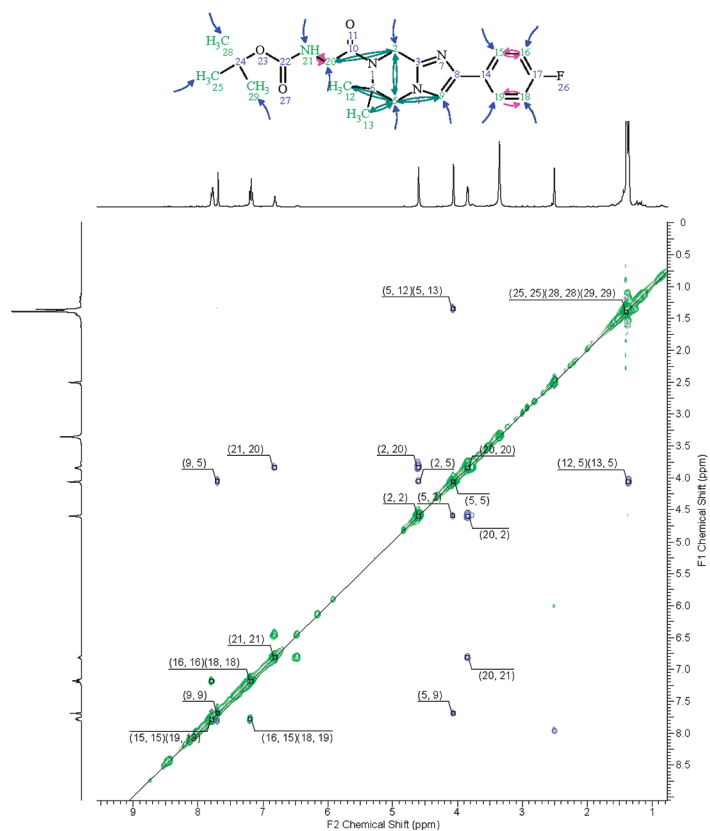
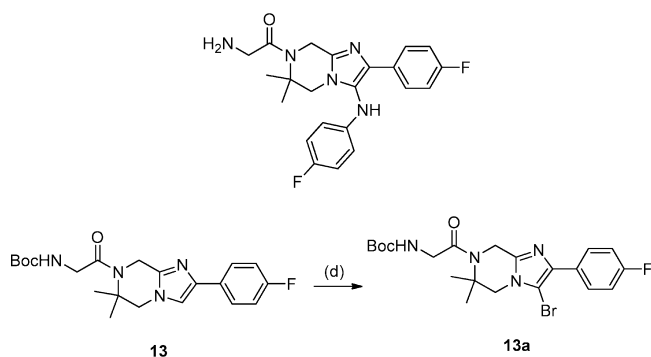


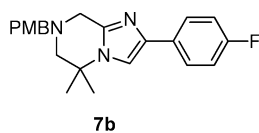


Table 9. continued

No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	2	2	4.6	4.6	15	15	16	7.79	7.19
2	5	2	4.07	4.59	16	16	16	7.18	7.18
3	20	2	3.85	4.6	17	18	18	7.18	7.18
4	2	5	4.6	4.05	18	19	18	7.79	7.19
5	5	5	4.06	4.06	19	18	19	7.2	7.78
6	9	5	7.7	4.05	20	19	19	7.79	7.79
7	12	5	1.37	4.06	21	2	20	4.6	3.83
8	13	5	1.37	4.06	22	20	20	3.85	3.85
9	5	9	4.06	7.69	23	21	20	6.82	3.84
10	9	9	7.69	7.69	24	20	21	3.85	6.81
11	5	12	4.07	1.35	25	21	21	6.81	6.81
12	5	13	4.07	1.35	26	25	25	1.39	1.39
13	15	15	7.79	7.79	27	28	28	1.39	1.39
14	16	15	7.2	7.78	28	29	29	1.39	1.39

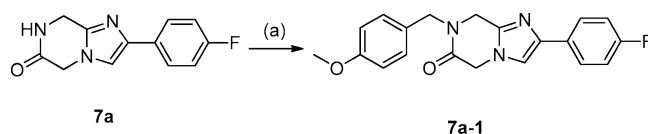


Example 25 was obtained from 13a by a  $\text{Pd}_2(\text{dba})_3$  mediated amination reaction with 4-fluoroaniline followed by a TFA mediated deprotection by analogy to compound 22 (55% yield over 2 steps).  $^1\text{H}$  NMR (400 MHz,  $\text{DMSO}-d_6$ ):  $\delta$  8.06 (m, 2H), 7.98 (m, 1H), 7.80 (m, 2H), 7.16 (m, 2H), 6.98 (m, 2H), 6.60 (m, 2H), 4.64 (m, 2H), 4.02 (m, 2H), 3.80 (m, 2H), 1.38 (s, 6H).  $m/z = 412.1$  ( $M + 1$ ).

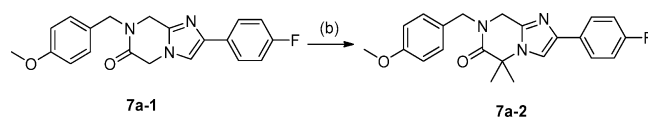


**Synthesis of 7b.** 2-(4-Fluorophenyl)-7-(4-methoxybenzyl)-5,5-dimethyl-5,6,7,8-tetrahydroimidazo[1,2-a]pyrazine:

Compound 7b was prepared from compound 7a ( $R_1 = R_2 = \text{H}$ ) by the following way:

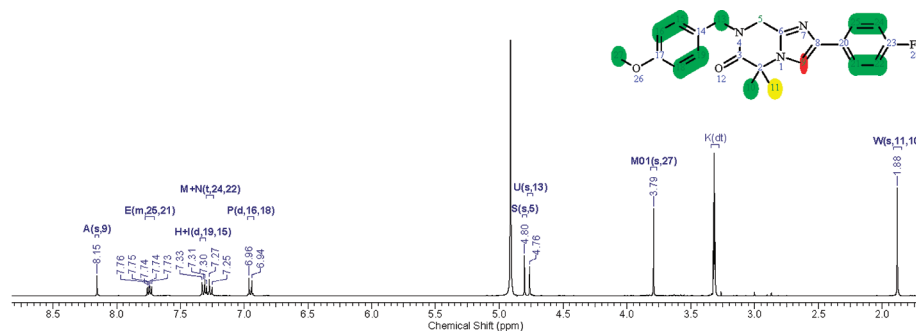


To a solution of compound 7a (231 mg, 1.0 mmol) in DMF (10 mL) were added KOH (168 mg, 3.0 mmol) and PMBCl (405  $\mu\text{L}$ , 3.0 mmol) at 0 °C. The reaction mixture was stirred at the same temperature for 2 h and at rt for 2 additional hours. The reaction mixture was purified by reverse phase HPLC. The HPLC fraction were evaporated under reduced pressure and neutralized by satd  $\text{NaHCO}_3$ . The compound was extracted by ethyl acetate and dried over sodium sulfate. The organic layer was concentrated under reduced pressure to give compound 7a-1 (221 mg, 63%) as a white solid.  $^1\text{H}$  NMR (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.67–7.63 (m, 2H), 7.28–7.25 (m, 2H), 7.07–7.02 (m, 2H), 6.89–6.86 (m, 2H), 4.74 (s, 2H), 4.70 (m, 2H), 4.53 (s, 2H), 3.79 (s, 3H).

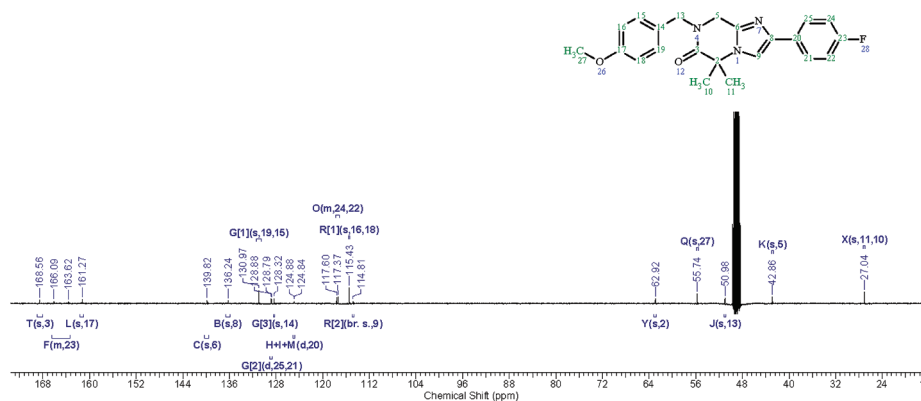


To a solution of compound 7a-1 (253 mg, 0.72 mmol) in DMF (15 mL) were added 60% sodium hydride (87 mg, 0.085 mmol) and methyl iodide (0.45 mL, 7.2 mmol) at rt. The reaction mixture was stirred at rt for 2 h. The reaction mixture was quenched with methanol and directly subjected to mass-triggered HPLC purification to give 7a-2 as a white solid after evaporation of

Table 10. Structure of 7a-2 Verified by 2-D NMR Techniques

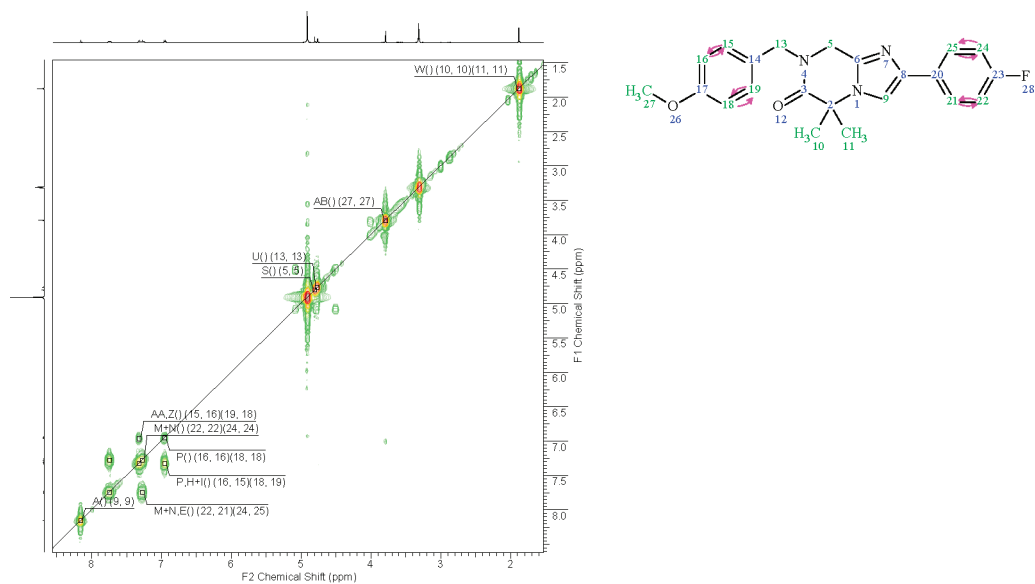


No.	Atom	Exp. Shift (ppm)	Calc. Shift (ppm)	Difference (ppm)	Multiplet	No.	Atom	Exp. Shift (ppm)	Calc. Shift (ppm)	Difference (ppm)	Multiplet
1	5	4.8	-	-	S	9	16	6.95	6.83	0.121	P
2	5	-	4.176	-	-	10	18	6.95	6.83	0.121	P
3	5	-	4.176	-	-	11	19	7.32	7.248	0.071	H+I
4	9	8.15	6.742	1.412	A	12	21	7.74	7.46	0.283	E
5	10	1.88	1.798	0.084	W	13	22	7.27	7.039	0.234	M+N
6	11	1.88	1.421	0.461	W	14	24	7.27	7.039	0.234	M+N
7	13	4.76	4.62	0.143	U	15	25	7.74	7.46	0.283	E
8	15	7.32	7.248	0.071	H+I	16	27	3.79	3.83	0.038	M01



No.	Atom	Exp. Shift (ppm)	Multiplet	No.	Atom	Exp. Shift (ppm)	Multiplet
1	2	62.92	Y	12	16	115.43	R[1]
2	3	168.56	T	13	17	161.27	L
3	5	42.86	K	14	18	115.43	R[1]
4	6	139.82	C	15	19	130.97	G[1]
5	8	136.24	B	16	20	124.86	H+I+M
6	9	114.81	R[2]	17	21	128.84	G[2]
7	10	27.04	X	18	22	117.48	O
8	11	27.04	X	19	23	164.86	F
9	13	50.98	J	20	24	117.48	O
10	14	128.32	G[3]	21	25	128.84	G[2]
11	15	130.97	G[1]	22	27	55.74	Q

Table 10. continued



No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	5	5	4.8	4.8	12	18	19	6.95	7.32
2	9	9	8.15	8.15	13	19	19	7.32	7.32
3	10	10	1.88	1.88	14	21	21	7.74	7.74
4	11	11	1.88	1.88	15	22	21	7.27	7.74
5	13	13	4.76	4.76	16	21	22	7.74	7.27
6	15	15	7.32	7.32	17	22	22	7.27	7.27
7	16	15	6.95	7.32	18	24	24	7.27	7.27
8	15	16	7.32	6.96	19	25	24	7.74	7.27
9	16	16	6.95	6.95	20	24	25	7.27	7.74
10	18	18	6.95	6.95	21	25	25	7.74	7.74
11	19	18	7.32	6.96	22	27	27	3.79	3.8

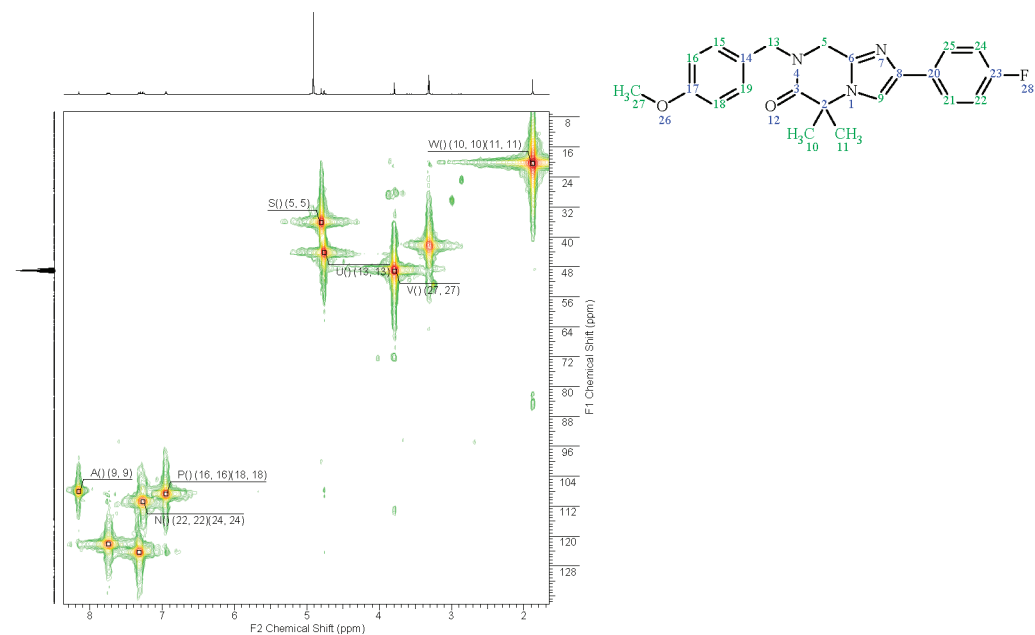
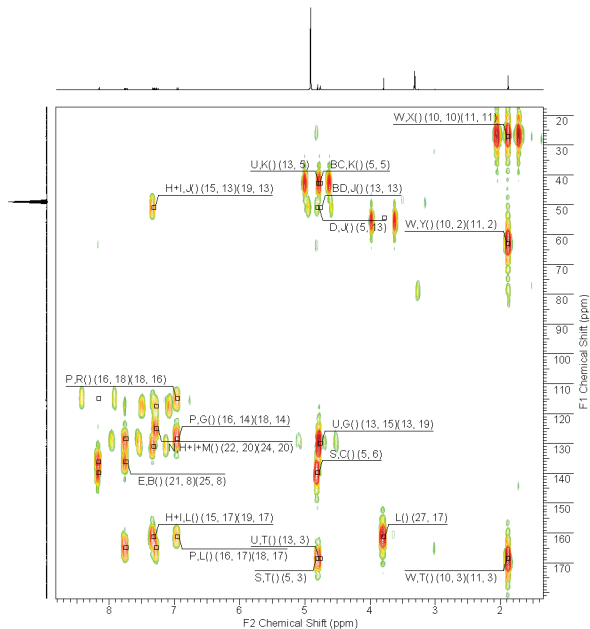
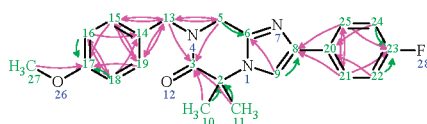


Table 10. continued

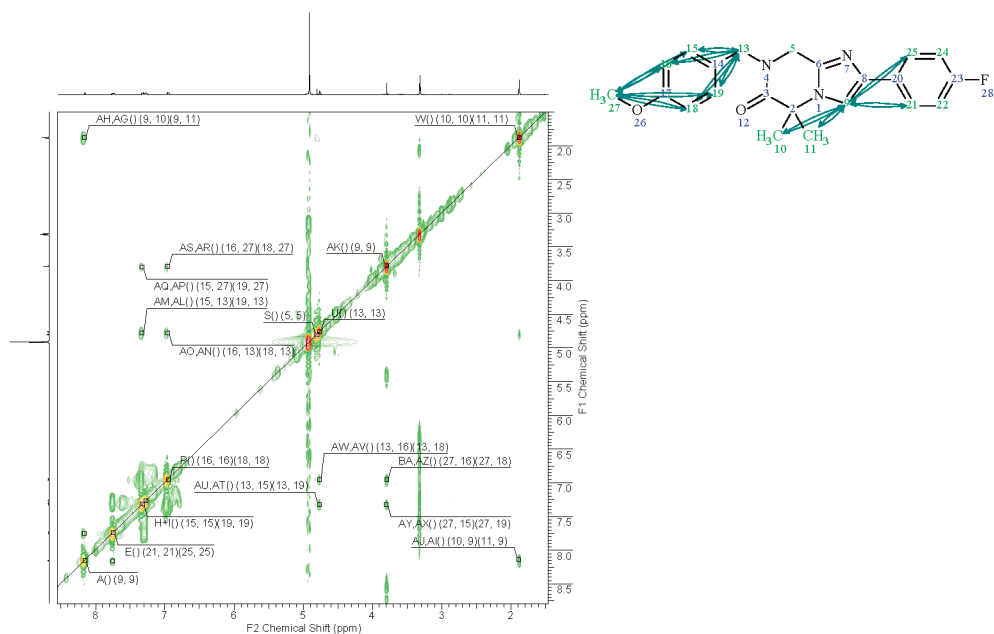
No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	5	5	4.8	36.13	9	18	18	6.95	108.76
2	9	9	8.15	108.09	10	19	19	7.32	124.31
3	10	10	1.88	20.31	11	19	19	7.32	124.33
4	11	11	1.88	20.31	12	21	21	7.74	122.18
5	13	13	4.76	44.27	13	22	22	7.27	110.83
6	15	15	7.32	124.31	14	24	24	7.27	110.83
7	15	15	7.32	124.33	15	25	25	7.74	122.18
8	16	16	6.95	108.76	16	27	27	3.79	49.12



No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	10	2	1.88	62.92	10	9	6	8.16	139.82
2	11	2	1.88	62.92	11	9	8	8.16	136.24
3	5	3	4.8	168.56	12	21	8	7.74	136.24
4	10	3	1.88	168.56	13	25	8	7.74	136.24
5	11	3	1.88	168.56	14	9	9	8.16	114.81
6	13	3	4.76	168.56	15	10	10	1.88	27.04
7	5	5	4.79	42.86	16	11	11	1.88	27.04
8	13	5	4.76	42.86	17	5	13	4.8	50.98
9	5	6	4.8	139.82	18	13	13	4.76	50.98

Table 10. continued

No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
19	15	13	7.32	50.98	34	15	19	7.32	130.97
20	19	13	7.32	50.98	35	19	19	7.32	130.97
21	16	14	6.95	128.32	36	22	20	7.27	124.86
22	18	14	6.95	128.32	37	24	20	7.27	124.86
23	13	15	4.76	129.84	38	21	21	7.74	128.32
24	15	15	7.32	130.97	39	25	21	7.74	128.32
25	19	15	7.32	130.97	40	22	22	7.27	117.48
26	18	16	6.95	114.81	41	21	23	7.74	164.86
27	15	17	7.32	161.27	42	22	23	7.27	164.86
28	16	17	6.95	161.27	43	24	23	7.27	164.86
29	18	17	6.95	161.27	44	25	23	7.74	164.86
30	19	17	7.32	161.27	45	24	24	7.27	117.48
31	27	17	3.79	161.27	46	21	25	7.74	128.32
32	16	18	6.95	114.81	47	25	25	7.74	128.32
33	13	19	4.76	129.84	48	27	27	3.78	54.41

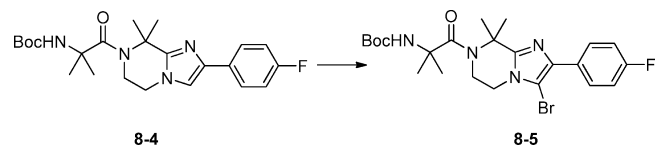
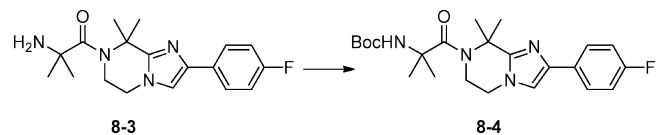
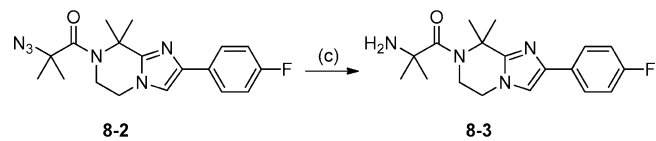
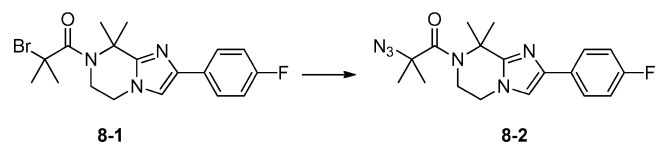
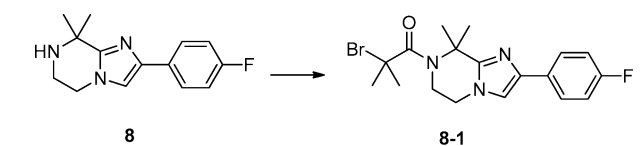
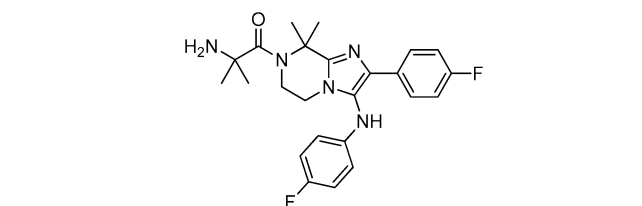
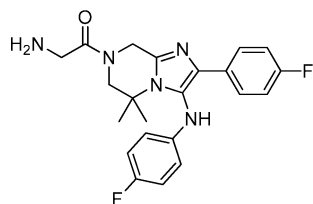
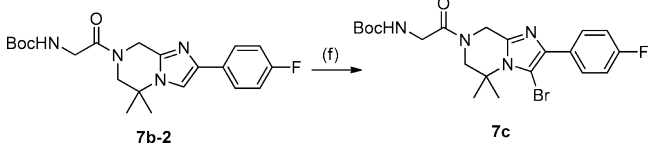
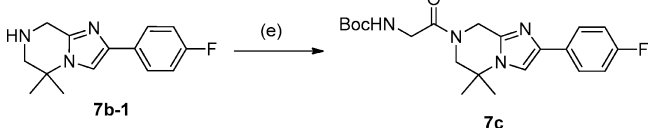
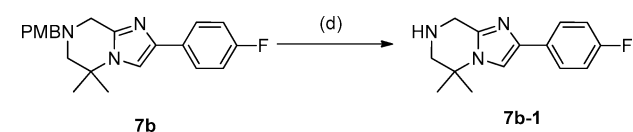
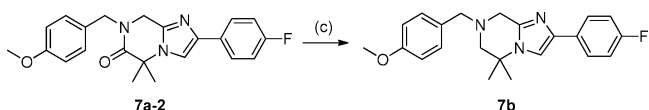


No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
1	5	5	4.8	4.8	11	11	11	1.88	1.88
2	9	9	3.8	3.79	12	13	13	4.76	4.76
3	9	9	8.15	8.15	13	15	13	7.33	4.78
4	10	9	1.88	8.14	14	16	13	6.96	4.78
5	11	9	1.88	8.14	15	18	13	6.96	4.78
6	21	9	7.76	8.16	16	19	13	7.33	4.78
7	25	9	7.76	8.16	17	13	15	4.77	7.32
8	9	10	8.16	1.88	18	15	15	7.32	7.32
9	10	10	1.88	1.88	19	27	15	3.8	7.32
10	9	11	8.16	1.88	20	13	16	4.77	6.96



Table 10. continued

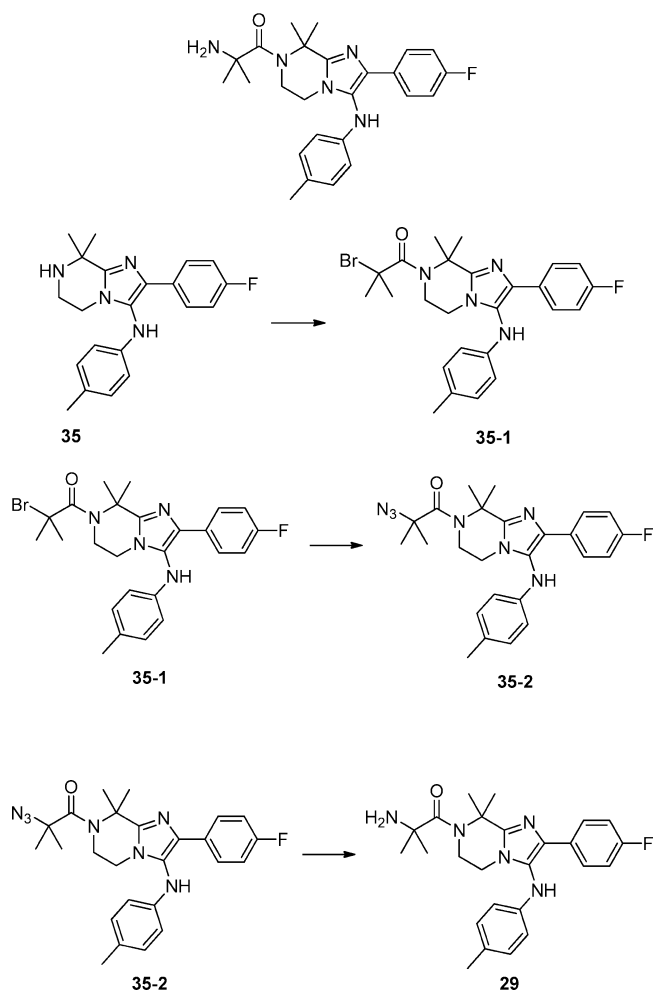
No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)	No.	F2 Atom	F1 Atom	F2 (ppm)	F1 (ppm)
21	16	16	6.95	6.95	30	21	21	7.74	7.74
22	27	16	3.8	6.96	31	22	22	7.27	7.27
23	13	18	4.77	6.96	32	24	24	7.27	7.27
24	18	18	6.95	6.95	33	9	25	8.16	7.75
25	27	18	3.8	6.96	34	25	25	7.74	7.74
26	13	19	4.77	7.32	35	15	27	7.33	3.8
27	19	19	7.32	7.32	36	16	27	6.96	3.8
28	27	19	3.8	7.32	37	18	27	6.96	3.8
29	9	21	8.16	7.75	38	19	27	7.33	3.8



acetonitrile–water in vacuo. The TFA salt was subsequently neutralized by saturated sodium bicarbonate solution and extracted with dichloromethane to get the compound as a free base. The organic layer was dried over sodium sulfate and

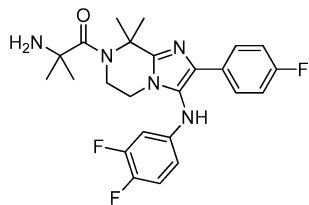
concentrated under reduced pressure to yield the desired compound **7a-2** (234 mg, 86%). The structure of **7a-2** was verified by 2-D NMR techniques (Table 10).

**Synthesis of Compound 7b.** To a solution of compound **7a-2** (170 mg, 0.45 mmol) in THF (9 mL) was added 1.0 N  $\text{BH}_3 \cdot \text{THF}$  (2.70 mL, 2.70 mmol) at rt. The reaction mixture was stirred at reflux for 2 h.



Pd/C was added (gas generated). The reaction mixture was stirred for 1 h. Solid was filtered off, and solvent was removed. The structure of crude compound **7b** was verified by proton NMR. The product was assumed to be of 100% yield and used in the next step without further purification. <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD) δ 8.1 (s, 1H), 7.74–7.70 (m, 2H), 7.34 (d, *J* = 8.6 Hz, 2H), 7.27 (m, 2H), 6.94 (d, *J* = 8.6 Hz, 2H), 3.91 (s, 2H), 3.80 (s, 3H), 3.79 (s, 2H), 2.87 (s, 2H), 1.61 (s, 6H).

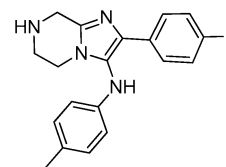
7-(4-Methoxybenzyl)-2-(4-fluorophenyl)-5,5-dimethyl-5,6,7,8-tetrahydroimidazo[1,2-*a*]pyrazine (1.07 g, 2.93 mmol, 1.00 equiv) (**7b**) was added to trifluoroacetic acid (15 mL). The resulting solution was stirred for 2 h at 70 °C in an oil bath. The resulting mixture was concentrated under vacuum. The residue was dissolved in 200 mL of dichloromethane. The pH value of the solution was adjusted to 8 with



aqueous sodium bicarbonate. **7b-1** was extracted from the aqueous layer using 3 × 100 mL of ethyl acetate. The mixture was dried over sodium sulfate and concentrated under vacuum. This resulted in **7b-1** (718 mg, quantitative) as a yellow solid. *m/z* = 246 (*M* + *H*).

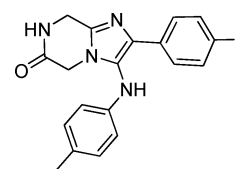
To a solution of **7b-1** (750 mg, 3.06 mmol, 1.00 equiv) in *N,N*-dimethylformamide (30 mL) were added 2-(*tert*-butoxycarbonyl)acetic acid (850 mg, 4.86 mmol, 1.50 equiv), HATU (1.3 g, 3.42 mmol, 1.10 equiv), and triethylamine (650 mg, 6.44 mmol, 2.00 equiv). The

resulting solution was stirred overnight at rt. The residue was dissolved in ethyl acetate (200 mL). The resulting mixture was washed with brine (3 × 100 mL), dried over anhydrous sodium sulfate, and

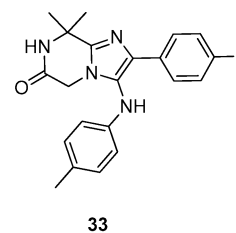


concentrated under vacuum. The residue was applied onto a silica gel column with petroleum ether/ethyl acetate (4:1). This resulted in 600 mg (49%) of **7c** as a pale-yellow solid. <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 7.75 (m, 2H), 7.22 (s, 1H), 7.13–7.07 (m, 2H), 4.99–4.87 (m, 2H), 4.12 (s, 2H), 3.91–3.65 (m, 2H), 1.57 (m, 6H), 1.43 (s, 9H).

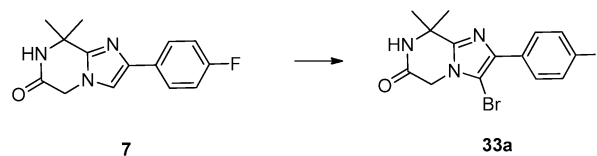
To a stirred solution of compound **7b-2** (583 g, 1.45 mmol) in dichloromethane (6 mL) was added Br<sub>2</sub> (82 μL, 1.60 mmol) in acetic acid (2 mL). The reaction mixture was stirred at rt for 30 min. Solvent was removed via rotavap at a temperature no higher than 20 °C. After neutralization, the residue was subjected to flash



chromatography (40 g, 0–100% ethyl acetate in hexane, 50 min, dry loading) purification to give 628 mg (90%) of the title compound **7c** as a yellow solid. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>) δ 7.81–7.78 (m, 2H), 7.07 (t, *J* = 8.6 Hz, 2H), 5.28 (s, NH), 4.84–4.7 (m, 2H), 4.07–3.58 (m, 4H), 1.49–1.44 (m, 6H), 1.28 (s, 9H).



33

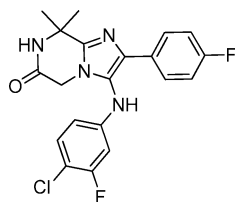


**Synthesis of 26.** 2-Amino-1-(2-(4-fluorophenyl)-3-(4-fluorophenylamino)-5,5-dimethyl-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8H)-yl)-2-ethanone:

Compound **26** was prepared from compound **7c** by a Pd<sub>2</sub>(dba)<sub>3</sub> mediated amination reaction with 4-fluoroaniline followed by a TFA mediated deprotection in 46% yield over 2 steps. <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD): δ 7.57 (m, 2H), 7.02 (m, 2H), 6.77 (m, 2H), 6.52 (m, 2H), 4.97–4.91 (m, 2H), 4.09–3.74 (m, 4H), 1.56 (s, 3H), 1.51 (s, 3H). *m/z* = 412.2 (*M* + 1).

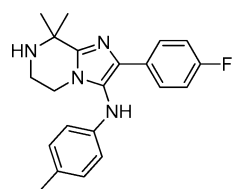
**Synthesis of 27.** 2-Amino-1-(2-(4-fluorophenyl)-3-((4-fluorophenylamino)-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8H)-yl)-2-methylpropan-1-one. The synthesis has been reported in the previous communication.<sup>5</sup>

**Synthesis of 28.** 2-Amino-1-(2-(4-fluorophenyl)-3-(4-fluorophenylamino)-8,8-dimethyl-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8H)-yl)-2-methylpropan-1-one:



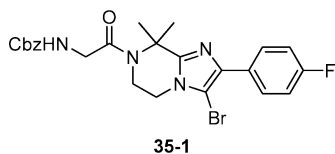
Compound **28** was prepared from compound **8** ( $R_1 = R_2 = \text{Me}$ ) in the following fashion using a slightly modified route to the one described in Scheme 2.

To a stirred solution of compound **8** (1.48 g, 6.04 mmol) and  $\text{NEt}_3$  (6.0 g, 59.4 mmol) in dichloromethane (20 mL) was added 2-bromo-2-methylpropanoyl bromide (14 g, 60.9 mmol) dropwise at rt. After being stirred for 3 h at rt, the reaction was then quenched by the addition of water (30 mL). The resulting solution was extracted with

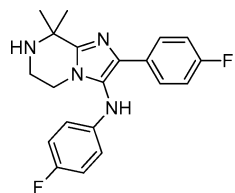


$3 \times 30$  mL of ethyl acetate. The combined organic layers were washed with brine, dried over anhydrous sodium sulfate, and concentrated under vacuum. This resulted in crude product as a dark solid, which was washed with EtOAc:petroleum ether (1:10) to remove the impurities to produce compound **8-1** as a gray solid.  $m/z$  396 ( $M + H$ ).

To a solution of compound **8-1** (2.0 g, 5.08 mmol, 1.00 equiv) in DMF (10 mL) was  $\text{NaN}_3$  (1.0 g, 15.38 mmol, 3.00 equiv) at rt. The



reaction mixture was stirred overnight at rt. The resulting solution was diluted with 300 mL of ethyl acetate. The resulting mixture was washed with  $3 \times 20$  mL of brine, dried over anhydrous sodium sulfate,



and concentrated under vacuum. The residue was applied onto a silica gel column with petroleum ether/EtOAc (5:1) to give compound **8-2** as a white solid.  $m/z$  357 ( $M + H$ ).

To a stirred solution of compound **8-2** (1.2 g, 3.37 mmol, 1.00 equiv) in methanol (20 mL) was added Pd/C (80 mg, 0.75 mmol, 0.20 equiv) at rt. The reaction mixture was evacuated and backfilled with  $\text{H}_2$ . The reaction mixture was stirred overnight at rt. The solids were filtered out. The resulting mixture was concentrated under vacuum. The solid was washed with petroleum ether. This resulted in compound **8-3** as a white solid.  $m/z$  331 ( $M + H$ ).

To a stirred solution of compound **8-3** (910 mg, 2.76 mmol, 1.00 equiv) in THF (50 mL) was added BOC anhydride (3.2 g, 14.68 mmol, 5.00 equiv), followed by aqueous NaOH (1N, 6 mL, 2.0 equiv) at rt. The resulting solution was stirred for 24 h at 40 °C. The resulting solution was concentrated under vacuum. The mixture was diluted with EtOAc (60 mL). The organic layer was washed with  $3 \times 10$  mL of brine, dried over  $\text{Na}_2\text{SO}_4$ , and concentrated under vacuum. The solid was collected by filtration and washed with *n*-hexane (5 mL) to give compound **8-4** as a white solid.  $^1\text{H NMR}$  (300 MHz,  $\text{CDCl}_3$ )  $\delta$

7.75–7.71 (m, 2H), 7.09–7.02 (m, 3H), 4.84(s, NH), 4.09–4.0 (m, 4H), 1.98 (s, 6H), 1.55 (s, 6H), 1.45 (s, 9H).

To a stirred solution of compound **8-4** (32 mg, 0.074 mmol) in dichloromethane (3 mL) was added  $\text{Br}_2$  (4.2  $\mu\text{L}$ , 0.082 mmol) in acetic acid (1 mL). The reaction mixture was stirred at rt for 30 min. Solvent was removed via rotavap at a temperature no higher than 20 °C. After neutralization, the residue was subjected to flash chromatography (4 g, 0–60% ethyl acetate in hexane, 16 min) purification to give the compound **8-5** as colorless oil, was collected by filtration, and washed with *n*-hexane (5 mL) to give compound **8-4** as a white solid.  $^1\text{H NMR}$  (400 MHz,  $\text{CDCl}_3$ )  $\delta$  7.91–7.88 (m, 2H), 7.10–7.04 (m, 2H), 5.11(s, NH), 4.01 (m, 2H), 3.96 (s, 2H), 1.94 (s, 6H), 1.51 (s, 6H), 1.41 (s, 9H).

Compound **28** was prepared from compound **8-5** by a  $\text{Pd}_2(\text{dba})_3$  mediated amination reaction with 4-fluoroaniline followed by a TFA mediated deprotection.  $^1\text{H NMR}$  (400 MHz,  $\text{MeOH}-d_4$ )  $\delta$  7.66 (dd,  $J = 9.2, 5.2$  Hz, 2H), 7.11 (t,  $J = 8.8$  Hz, 2H), 6.81 (t,  $J = 8.8$  Hz, 2H), 6.72–6.68 (m, 2H), 4.04–3.99 (m, 4H), 2.01 (s, 6H), 1.66 (s, 6H).  $m/z = 440.1$  ( $M + 1$ ).

**Synthesis of 29.** 2-Amino-1-(2-(4-fluorophenyl)-8,8-dimethyl-3-(*p*-tolylamino)-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8H)-yl)-2-methylpropan-1-one:

Compound **29** was prepared from **35** by the following way:

To a stirred solution of **35** (21 mg, 0.06 mmol) and  $\text{Et}_3\text{N}$  (83  $\mu\text{L}$ , 0.60 mmol) in dry dichloromethane (6 mL) was added 2-bromo-2-methylpropanoyl bromide (71  $\mu\text{L}$ , 0.60 mmol). The reaction mixture was stirred at rt for 5 h. The reaction mixture was concentrated and subjected to mass-triggered LC/MS purification directly. The obtained solution was concentrated to give 22 mg (73%) of compound **35-1** as yellow oil after neutralization.  $^1\text{H NMR}$  (400 MHz,  $\text{MeOH}-d_4$ )  $\delta$  7.71–7.66 (m, 2H), 7.23–7.19 (m, 2H), 7.02 (d,  $J = 8.4$  Hz, 2H), 6.66 (d,  $J = 8.4$  Hz, 2H), 4.31 (m, 2H), 4.09 (m, 2H), 2.21 (s, 3H), 2.06 (s, 6H), 2.00 (s, 6H).

To a solution of compound **35-1** (22 mg, 0.044 mmol) in DMF (3 mL) was added  $\text{NaN}_3$  (8.6 mg, 0.132 mmol) at rt. The reaction mixture was stirred at rt for 2 h. The reaction mixture was directly subjected to mass-triggered HPLC purification directly. The obtained MeCN/aqueous solution was combined and concentrated to give 15 mg (75%) of compound **35-2** as yellow oil after neutralization.  $^1\text{H NMR}$  (400 MHz,  $\text{MeOH}-d_4$ )  $\delta$  7.70–7.66 (m, 2H), 7.23–7.19 (m, 2H), 7.04 (d,  $J = 8.3$  Hz, 2H), 6.66 (d,  $J = 8.4$  Hz, 2H), 4.26 (m, 2H), 4.07–4.05 (m, 2H), 2.24 (s, 3H), 2.03 (s, 6H), 1.58 (s, 6H).

To a solution of compound **35-2** (15 mg, 0.033 mmol) in MeOH (3 mL) was added 10% Pd/C (4 mg, 0.003 mmol) at rt. Air was removed, and  $\text{H}_2$  was filled. The reaction mixture was stirred at rt for 2 h. Solid was filtered off, and solvent was removed. The reaction mixture was directly subjected to mass-triggered HPLC purification to give 15 mg (100%) of the title compound yellow oil.  $^1\text{H NMR}$  ( $\text{MeOH}-d_4$ , 400 Hz)  $\delta$  7.78–7.75 (m, 2H), 7.06–6.97 (m, 4H), 6.56 (d,  $J = 8.4$  Hz, 2H), 5.16 (s, NH), 4.33 (t,  $J = 4.8$  Hz, 2H), 3.74 (t,  $J = 4.8$  Hz, 2H), 2.29 (s, 3H), 1.95 (s, 6H), 1.45 (s, 6H).  $m/z = 436.2$  ( $M + 1$ ).

**Synthesis of 30.** 2-Amino-1-(3-(3,4-difluorophenylamino)-2-(4-fluorophenyl)-8,8-dimethyl-5,6-dihydroimidazo[1,2-*a*]pyrazin-7(8H)-yl)-2-methylpropan-1-one:

Compound **30** was prepared from compound **8-5** by a  $\text{Pd}_2(\text{dba})_3$  mediated amination reaction with 3,4-difluoroaniline followed by a TFA mediated deprotection.  $^1\text{H NMR}$  (400 MHz,  $\text{MeOH}-d_4$ )  $\delta$  7.63–7.60 (m, 2H), 7.14 (t,  $J = 8.4$  Hz, 2H), 7.01 (dd,  $J = 18.8, 8.8$  Hz, 1H), 6.68–6.63 (m, 1H), 6.5 (m, 1H), 4.03 (m, 2H), 4.01 (m, 2H), 2.01 (s, 6H), 1.66 (s, 6H).  $m/z = 458.1$  ( $M + 1$ ).

**Synthesis of 31.** 2-(4-Fluorophenyl)-*N*-(*p*-tolyl)-5,6,7,8-tetrahydroimidazo[1,2-*a*]pyrazin-3-amine:

The details of the synthesis involving an Ugi reaction between 4-fluorobenzaldehyde, 2-aminopyrazine, and 1-isocyanato-4-methylbenzene has been reported in the previous communication.<sup>5</sup>  $^1\text{H NMR}$  (400 MHz,  $\text{MeOH}-d_4$ )  $\delta$  7.73 (d,  $J = 6.8$  Hz, 2H), 7.00–6.99 (m, 4H), 6.49 (d,  $J = 8.0$  Hz, 2H), 4.01 (s, 2H), 3.72 (m, 2H), 3.15 (m, 2H), 2.19 (s, 3H).  $m/z = 324.2$  ( $M + 1$ ).

**Synthesis of 32.** 2-(4-Fluorophenyl)-3-(*p*-tolylamino)-7,8-dihydroimidazo[1,2-*a*]pyrazin-6(*SH*)-one:

Compound **32** was synthesized from compound **7** ( $R_1 = R_2 = H$ )  $Pd_2(dba)_3$  mediated coupling reaction with *p*-toluidine.  $^1H$  NMR (400 MHz, DMSO- $d_6$ )  $\delta$  8.55 (s, 1H), 7.77–7.75 (m, 2H), 7.26–7.14 (m, 4H), 7.01–6.95 (m, 2H), 6.53 (m, 1H), 4.53 (m, 2H), 4.22 (m, 2H), 2.17 (s, 3H).  $m/z = 337.2$  ( $M + 1$ ).

**Synthesis of 33.** 2-(4-Fluorophenyl)-8,8-dimethyl-3-(*p*-tolylamino)-7,8-dihydroimidazo[1,2-*a*]pyrazin-6(5H)-one:

Compound **33a** was prepared from **7** by the following way:

To a stirred solution of compound **7** (390 mg, 1.51 mmol, 1.00 equiv) in dichloromethane (20 mL) was added NBS (0.28 g, 1.00 equiv). The resulting solution was stirred for 2 h at rt. The solid was filtered out, and the mixture was washed with a saturated solution of  $Na_2S_2O_3$  and dried over  $Na_2SO_4$ . The mixture was concentrated under vacuum. The solids were purified by silica gel chromatography (petroleum ether/EtOAc = 1:2) to result in 432 mg (85%) of the title compound as a white solid.  $^1H$  NMR (300 MHz, DMSO- $d_6$ )  $\delta$  8.76 (s, 1H), 7.94–7.89 (m, 2H), 7.31–7.25 (m, 2H), 4.57 (s, 2H), 1.56 (s, 6H).  $m/z = 338$  ( $M + 1$ ).

Compound **33** was prepared from compound **33a** by a  $Pd_2(dba)_3$  mediated amination reaction with *p*-toluidine.  $^1H$  NMR (MeOH- $d_4$ , 400 Hz)  $\delta$  7.62–7.58 (m, 2H), 7.05 (t,  $J = 8.4$  Hz, 2H), 6.92 (d,  $J = 8.4$  Hz, 2H), 6.52 (d,  $J = 8.4$  Hz, 2H), 4.42 (s, 2H), 2.12 (s, 3H), 1.71 (s, 6H).  $m/z = 365.2$  ( $M + 1$ ).

**Synthesis of 34.** 3-(4-Chloro-3-fluorophenylamino)-2-(4-fluorophenyl)-8,8-dimethyl-7,8-dihydroimidazo[1,2-*a*]pyrazin-6(5H)-one:

Compound **34** was prepared from compound **7b** by a  $Pd_2(dba)_3$  mediated amination reaction with 4-chloro-3-fluoroaniline.  $^1H$  NMR (MeOH- $d_4$ , 400 Hz)  $\delta$  7.74–7.70 (m, 2H), 7.21 (t,  $J = 8.4$  Hz, 1H), 7.08 (m, 2H), 6.49–6.40 (m, 2H), 4.43 (s, 2H), 1.72 (s, 6H).  $m/z = 403.1$  ( $M + 1$ ).

**Synthesis of 35.** 2-(4-Fluorophenyl)-8,8-dimethyl-*N*-*p*-tolyl-5,6,7,8-tetrahydroimidazo[1,2-*a*]pyrazin-3-amine:

Compound **35** was prepared from compound **35-1** (shown below and prepared by analogy to compound **9** and using Cbz-glycine) by a  $Pd_2(dba)_3$  mediated amination reaction with *p*-toluidine followed by a 6N HCl mediated hydrolysis.  $^1H$  NMR (MeOH- $d_4$ , 400 Hz)  $\delta$  7.78–7.75 (m, 2H), 7.06–6.97 (m, 4H), 6.56 (d,  $J = 8.4$  Hz, 2H), 5.16 (s, NH), 4.33 (t,  $J = 4.8$  Hz, 2H), 3.74 (t,  $J = 4.8$  Hz, 2H), 2.29 (s, 3H), 1.95 (s, 6H).  $m/z = 436.2$  ( $M + 1$ ).

**Synthesis of 36.** *N*,2-Bis(4-fluorophenyl)-8,8-dimethyl-5,6,7,8-tetrahydroimidazo[1,2-*a*]pyrazin-3-amine:

Compound **36** was prepared from compound **35-1** by a  $Pd_2(dba)_3$  mediated amination reaction with 4-fluoroaniline followed by a 6N HCl mediated hydrolysis.  $^1H$  NMR (400 MHz, MeOH- $d_4$ )  $\delta$  7.62–7.58 (m, 2H), 7.12–7.08 (m, 2H), 6.68–6.63 (m, 2H), 6.67 (m, 2H), 4.16 (m, 2H), 3.78 (m, 2H), 1.92 (s, 6H).  $m/z = 355.1$  ( $M + 1$ ).

## ■ ASSOCIATED CONTENT

### ● Supporting Information

In vitro ADME of selected analogues; PAMPA vs Log *P*; Caco-2 (A–B) vs Log *P*; Caco-2 A–B/B–A ratio vs Log *P*; Caco-2 (B–A) vs Log *P*. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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### Author Contributions

#Equal contribution to this work

### Notes

The authors declare no competing financial interest.

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